

# Investigation of C<sub>18</sub> Bonding Chemistry on Novel Hybrid Organic/Inorganic Particles

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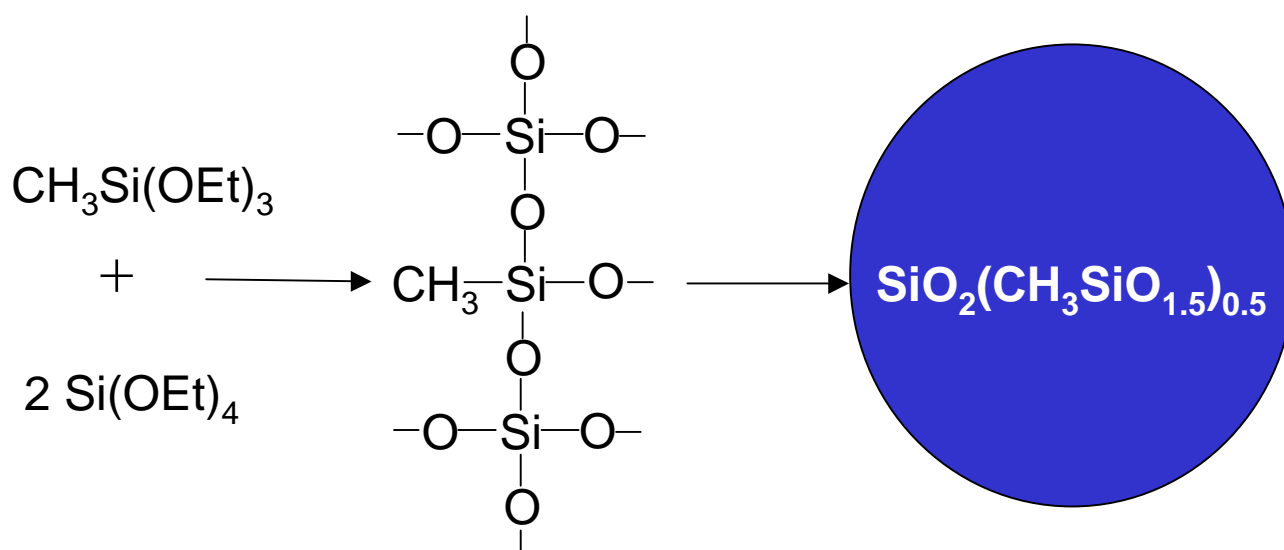
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## Summary:

A novel hybrid organic/inorganic particle is compared to a classical high purity silica with respect to C<sub>18</sub> silane bonding chemistry. Monofunctional, trifunctional, and embedded polar group C<sub>18</sub> silanes were bonded onto both materials under similar reaction conditions. Bonded phase data are presented that characterize the differences and similarities between the two substrates. In addition, chromatographic data are presented for each bonded phase, where capacity factor and peak shape were measured using a set of neutral, polar, and basic test analytes.

# Hybrid Organic/Inorganic Particle Technology

- Hybrid Organic/Inorganic materials contain both organic and inorganic components
- The hybrid particles described here were synthesized from  $\text{Si}(\text{OEt})_4$  and  $\text{CH}_3\text{Si}(\text{OEt})_3$ :



- Hybrid particles combine:
  - efficiency and mechanical strength of silica
  - extended pH range and absence of base tailing of organic polymers

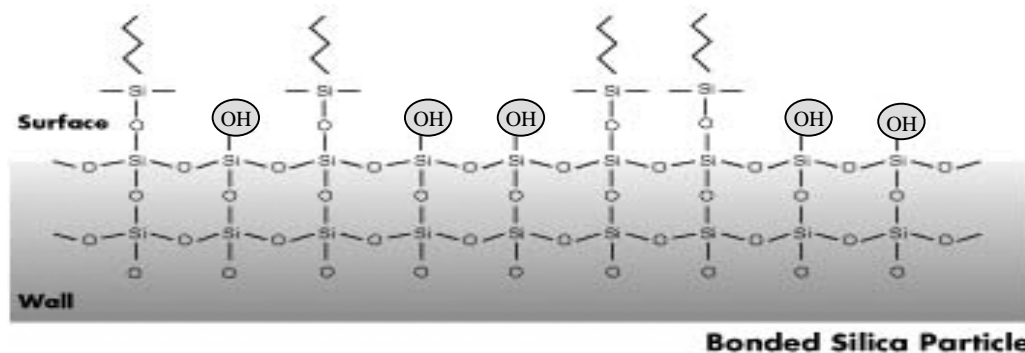
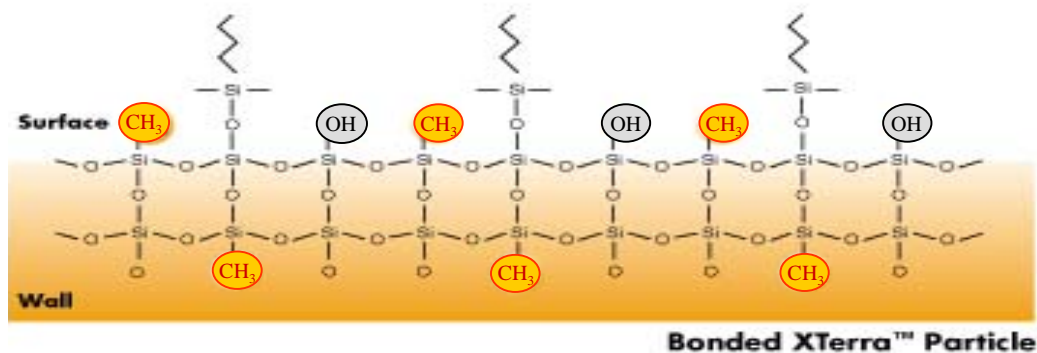
# Characteristics & Surface Structure: Hybrid vs. Silica Supports

## •Characteristics of Silica and Hybrid Supports:

	Hybrid*	Silica
Avg. Particle Diameter, $\mu\text{m}$	5.0	5.0
Specific Surface Area, $\text{m}^2/\text{g}$	173, 174	325
Pore Volume, $\text{cm}^3/\text{g}$	0.72, 0.69	0.84
Avg. Pore Diameter, $\text{\AA}$	145, 140	92
Carbon Content, % w/w	6.9, 6.6	0.0
Aluminum Content, ppm w/w	2, 1	0
Iron Content, ppm w/w	8, 3	4
Sodium Content, ppm w/w	4, 1	0

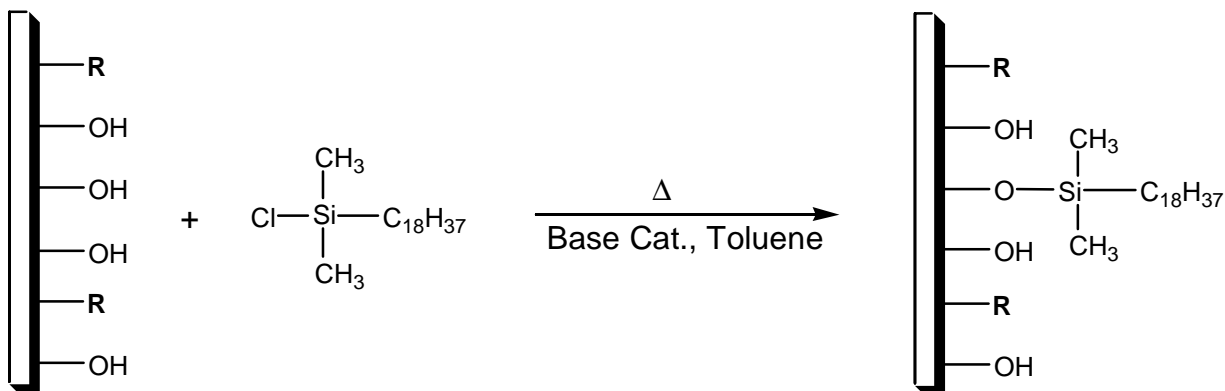
\* Two Hybrid Lots Used. Second entry used for Embedded Carbamate Bonding.

## •Surface Structure of Silica and Hybrid Supports:

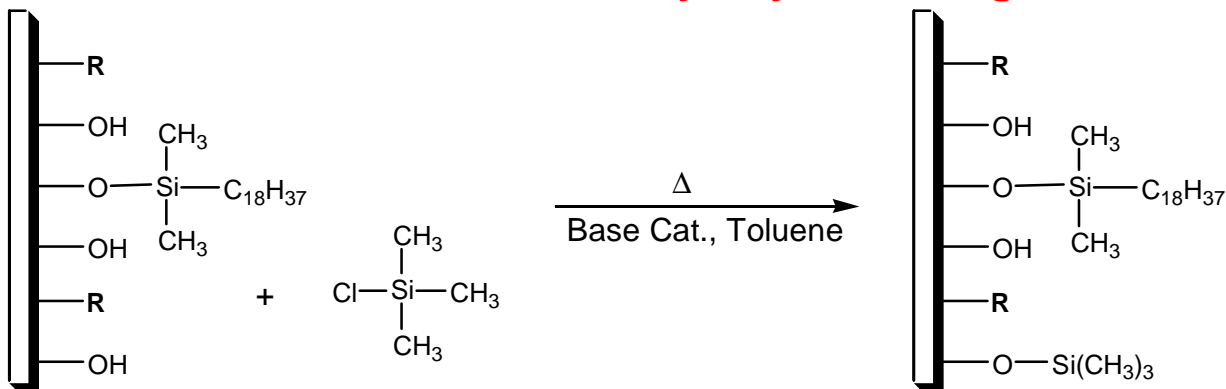


# Monofunctional C<sub>18</sub> Phase Data

## Surface Reaction #1: C<sub>18</sub> Bonding



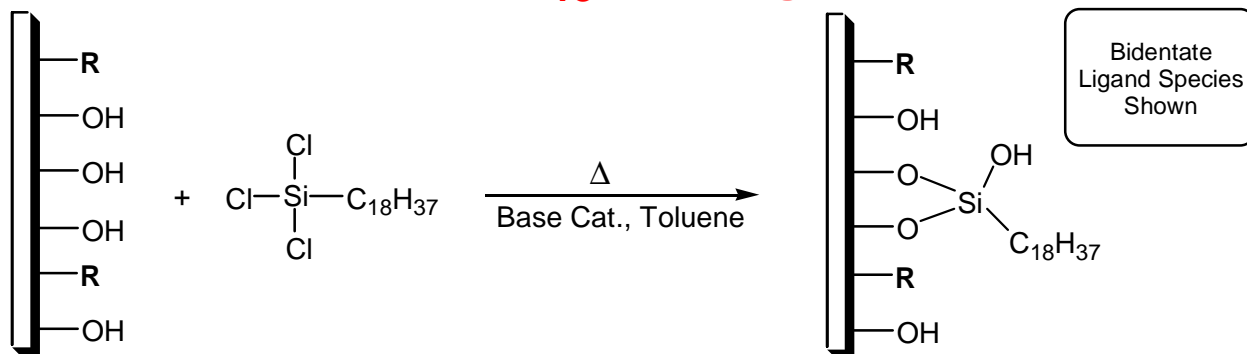
## Surface Reaction #2: Trimethylsilyl Bonding



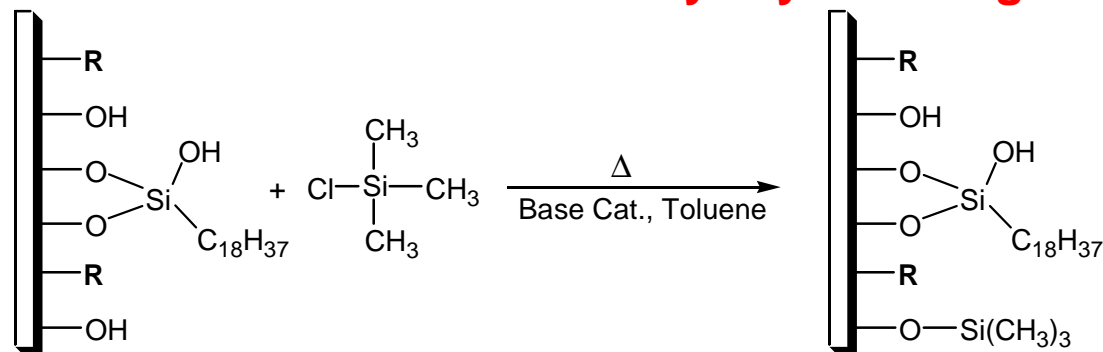
%C From:	Hybrid Phase R = CH <sub>3</sub>	Silica Phase R = OH
Base Particle	6.9	0.0
C <sub>18</sub> Ligand	8.0	19.2
TMS Ligand	0.3	0.3
Total %C	15.2	19.5
C <sub>18</sub> Surface Conc. (μmol/m <sup>2</sup> )	2.16	3.19

# Trifunctional C<sub>18</sub> Phase Data

## Surface Reaction #1: C<sub>18</sub> Bonding



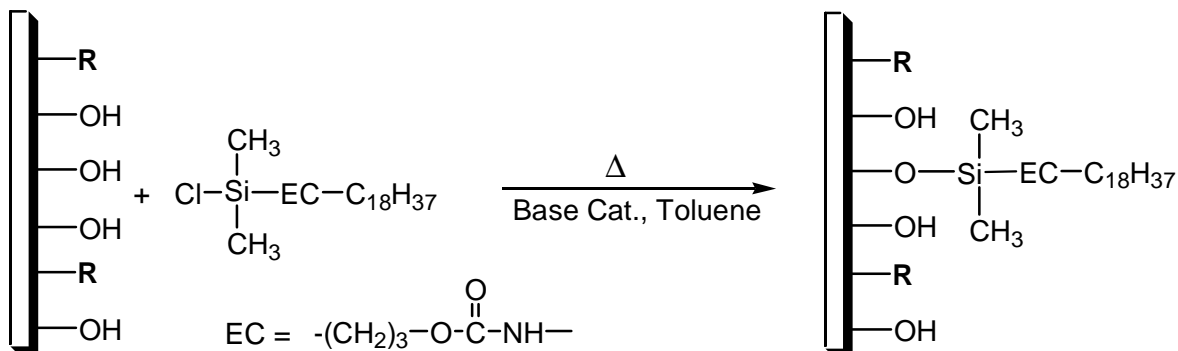
## Surface Reaction #2: Trimethylsilyl Bonding



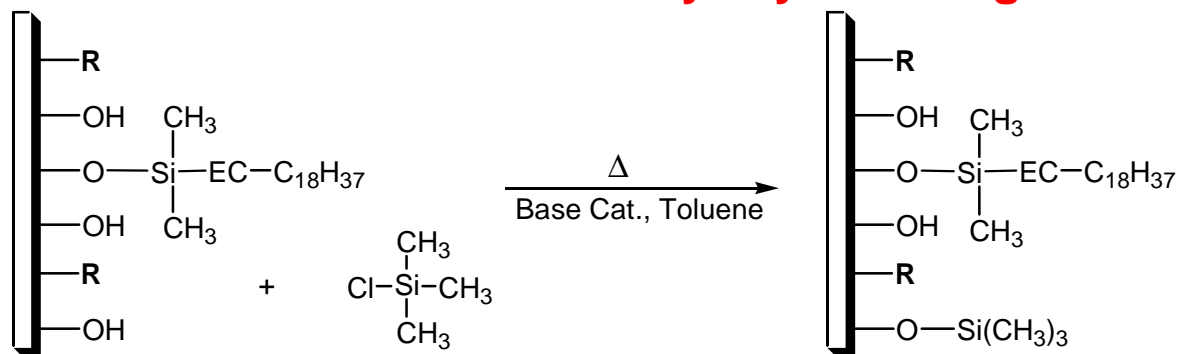
%C From:	Hybrid Phase R = CH <sub>3</sub>	Silica Phase R = OH
Base Particle	6.9	0.0
C <sub>18</sub> Ligand	8.0	18.4
TMS Ligand	0.8	0.8
Total %C	15.7	19.2
C <sub>18</sub> Surface Conc. (μmol/m <sup>2</sup> )	2.42	3.51

# Embedded Carbamate C<sub>18</sub> Phase Data

## Surface Reaction #1: Embedded Carbamate C<sub>18</sub> Bonding

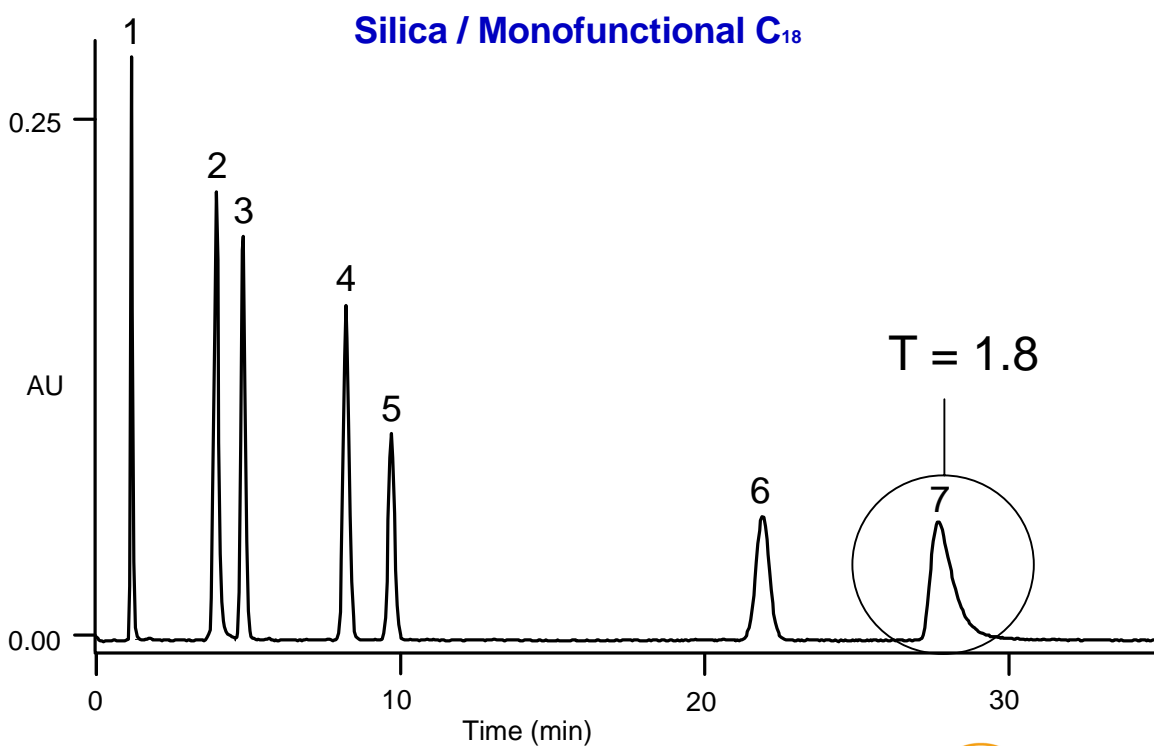
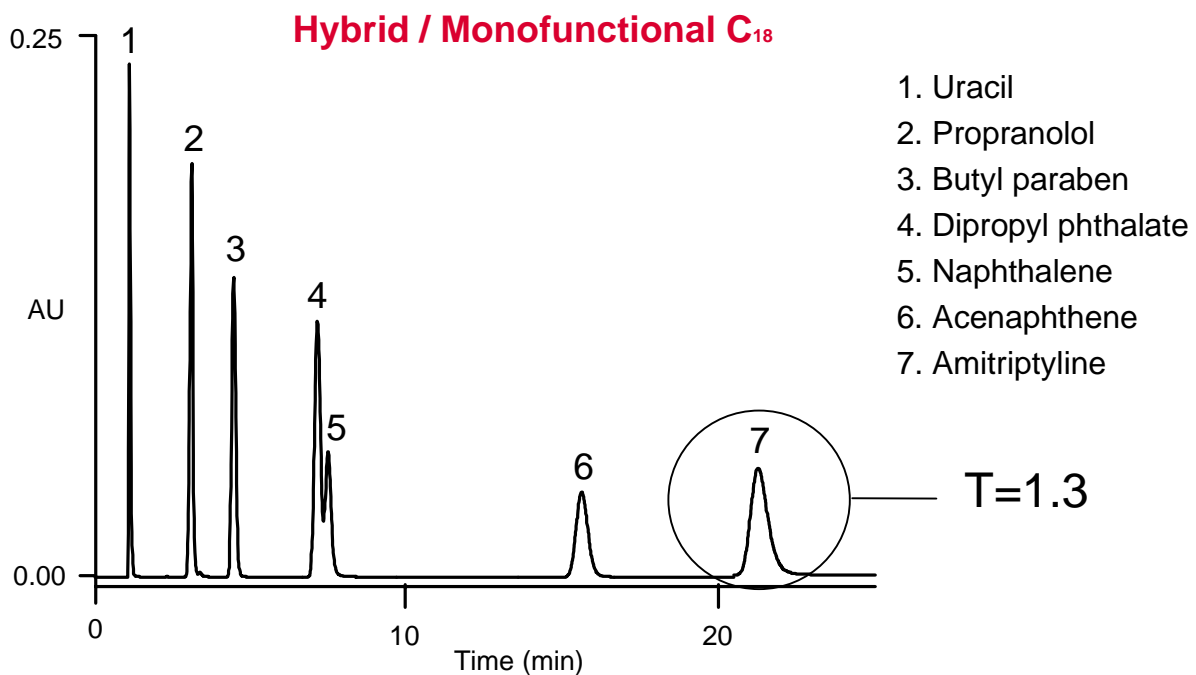


## Surface Reaction #2: Trimethylsilyl Bonding

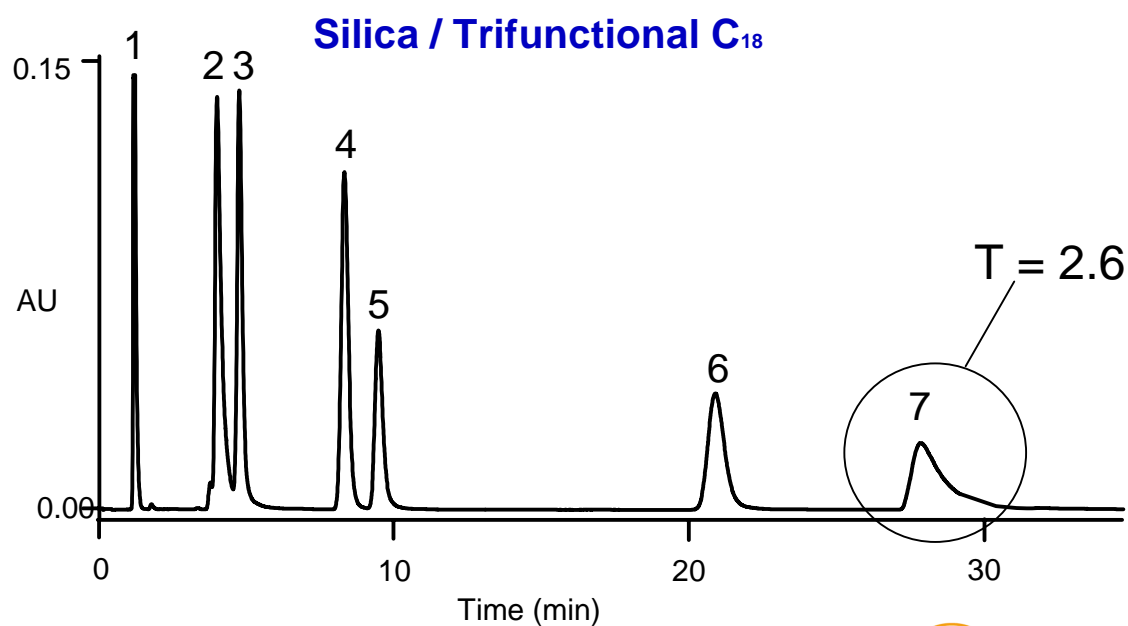
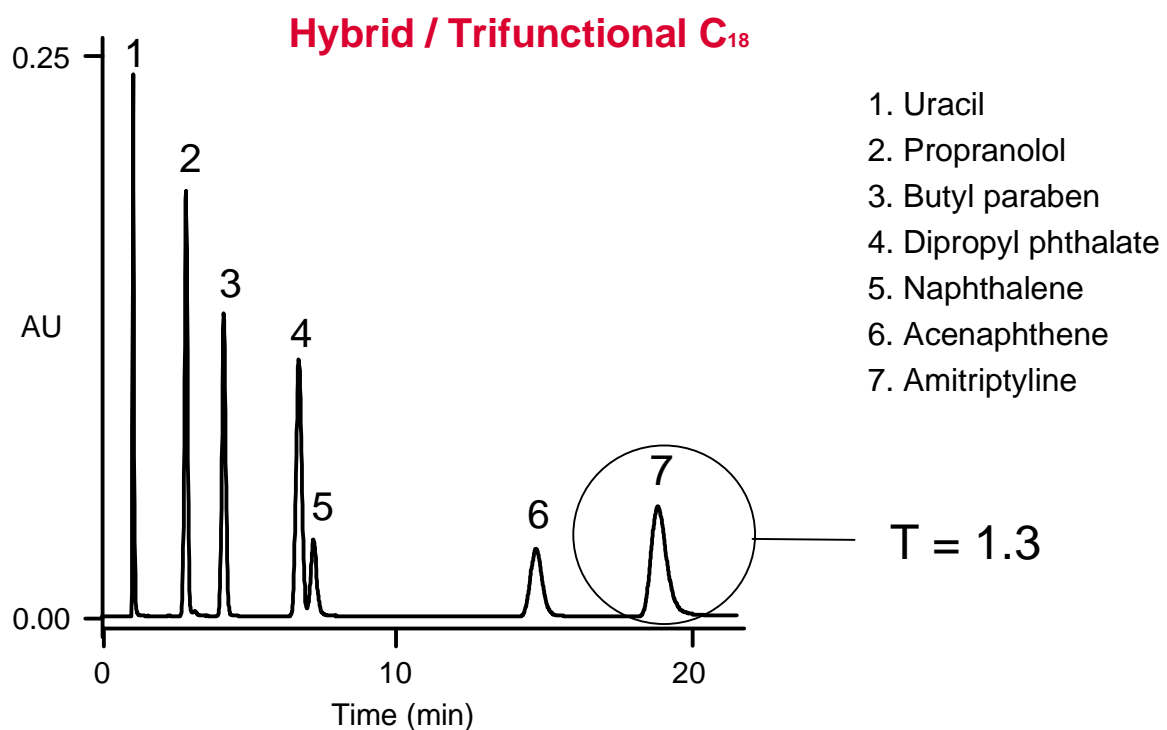


%C From:	Hybrid Phase R = CH <sub>3</sub>	Silica Phase R = OH
Base Particle	6.6	0.0
C <sub>18</sub> Ligand	8.1	21.2
TMS Ligand	0.1	0.3
Total %C	14.8	21.5
C <sub>18</sub> Surface Conc. (μmol/m <sup>2</sup> )	1.81	3.14

# Monofunctional C<sub>18</sub> Chromatograms

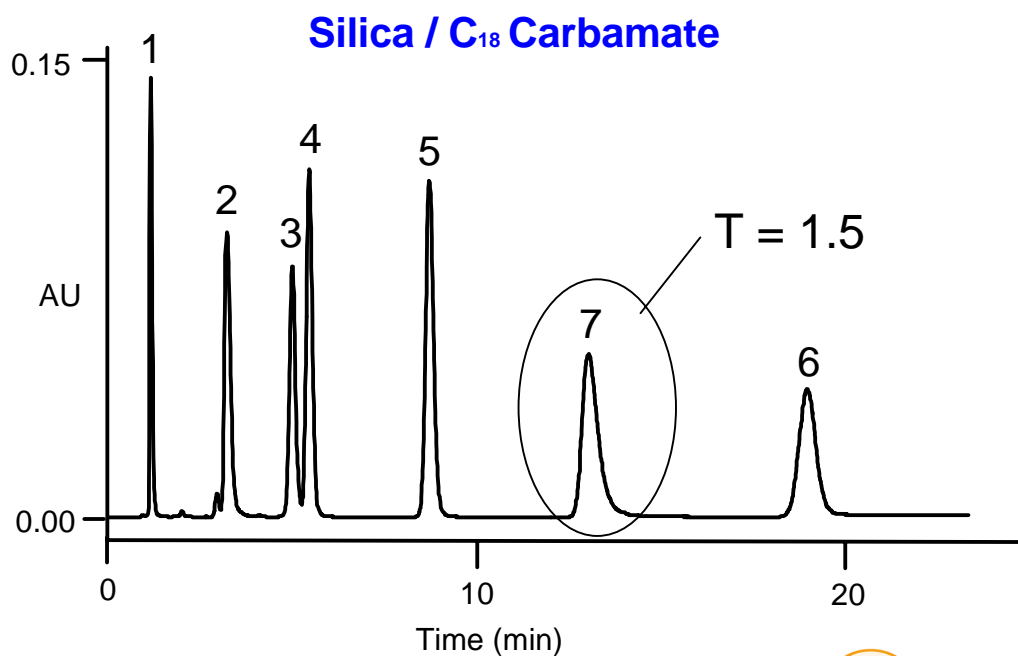
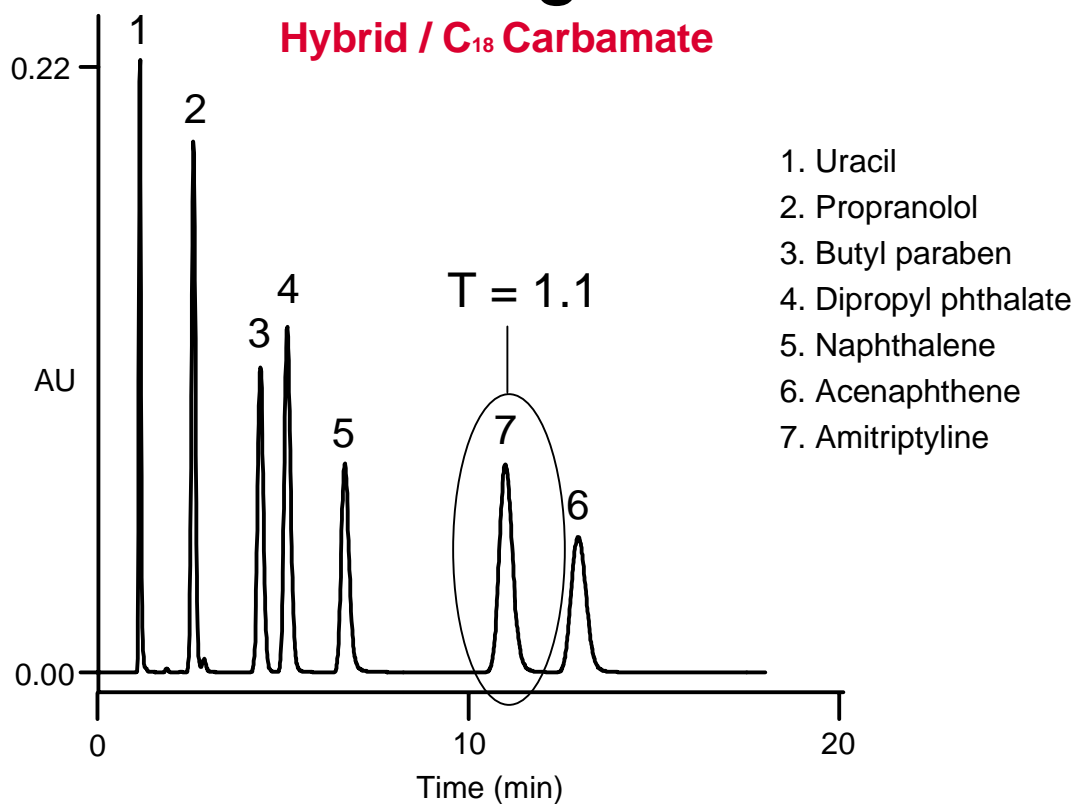


# Trifunctional C<sub>18</sub> Chromatograms





# Embedded Carbamate C<sub>18</sub> Chromatograms



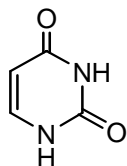
# Chromatographic Testing

## Conditions

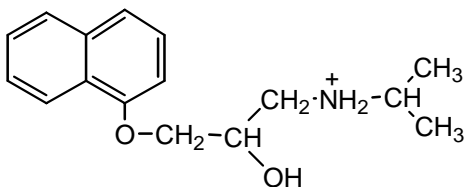
- 65:35 MeOH/20 mM  $\text{KH}_2\text{PO}_4/\text{K}_2\text{HPO}_4$ , pH 7.00
- Flow Rate: 1.0 mL/min
- Temperature:  $23.4 \pm 0.1$  °C

## Analytes

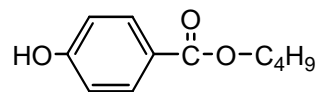
1) Uracil



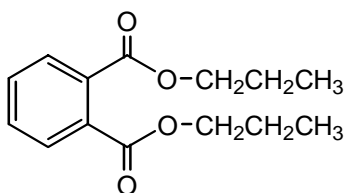
2) Propranolol,  $\text{pK}_a = 9.6$



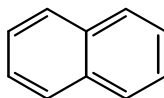
3) Butyl paraben



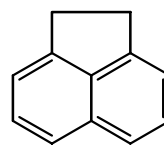
4) Dipropyl phthalate



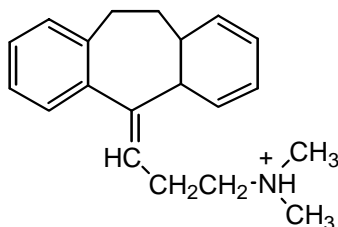
5) Naphthalene



6) Acenaphthene



7) Amitriptyline,  $\text{pK}_a = 9.4$



# Chromatographic Data

Ligand & Analyte	Hybrid Phase R = CH <sub>3</sub>	Silica Phase R = OH
<i>Monofunctional C<sub>18</sub></i>		
	<i>Retention Factor k (USP Tailing)</i>	
1. Uracil	-----	-----
2. Propranolol	1.64 (0.8)	2.01 (1.6)
3. Butyl paraben	2.74	2.67
4. Dipropyl phthalate	4.94	5.25
5. Naphthalene	5.22	6.37
6. Acenaphthene	11.90 (1.1)	15.64 (1.0)
7. Amitriptyline	16.54 (1.3)	20.02 (1.8)
<i>Trifunctional C<sub>18</sub></i>		
1. Uracil	-----	-----
2. Propranolol	1.55 (0.9)	2.28
3. Butyl Paraben	2.63	2.90
4. Dipropyl phthalate	4.81	5.79
5. Naphthalene	5.24	6.72
6. Acenaphthene	11.70 (1.1)	15.99 (1.3)
7. Amitriptyline	15.23 (1.3)	21.56 (2.6)
<i>Embedded Carbamate C<sub>18</sub></i>		
1. Uracil	-----	-----
2. Propranolol	1.17 (1.0)	1.78 (1.6)
3. Butyl Paraben	2.63	3.27
4. Dipropyl phthalate	3.22	3.67
5. Naphthalene	4.47	6.45
6. Acenaphthene	9.56 (1.1)	15.19 (1.1)
7. Amitriptyline	7.97 (1.1)	10.21 (1.5)

# Conclusions

## Hybrid vs. Silica Based C<sub>18</sub> Bonded Phases

- For all three ligands **lower C<sub>18</sub> surface concentrations** were obtained on the Hybrid particles due to their reduced surface silanol concentration
- **Hybrid C<sub>18</sub> Phases showed reduced analyte retentions** on all three phases due to the lower surface area of the particle and the lower C<sub>18</sub> surface concentrations
- **Hybrid C<sub>18</sub> Phases showed similar** efficiency and selectivity as the corresponding Silica phases
- **Hybrid C<sub>18</sub> Phases** showed reduced tailing factors for basic analytes, due to their reduced surface silanol concentration

*Note:*

*XTerra™ packings based on Hybrid Particle Technology are now available from Waters Corporation*