

# **MILLENNIUM APPLICATION**

## **BRIEF**

**TOPIC: Capillary Ion Analysis (CIA) of anions, cations, organic acids and metals in nickel plating baths**

**AUTHOR: Stuart Oehle**

**Date: 12/16/94**

### **Introduction:**

Analysis of nickel plating baths was done using a Q4000e CIA with data reduction done using the CIA option in the Millennium 2010 version 2.1 software. Analysis of anions and organic acids in a nickel plating solution was done simultaneously using either a chromate or phosphate electrolyte. Using the chromate electrolyte allows for all anions and organic acids to be resolved in less than 5 minutes. Using the phosphate electrolyte the analysis is accomplished in less than 7 minutes with a slightly improved peak shape for the citrate ion. Sulfate, however, is not detectable using the phosphate electrolyte. These two methods provide adequate analysis in a relatively fast time. Figures 1-3 are of a standard and sample, before and after plating, analyzed using the chromate electrolyte. Figures 4-6 are of a standard and another set of samples, before and after plating, analyzed using the phosphate electrolyte. Of main interest to this customer was monitoring the citrate levels during the plating process to evaluate when the citrate had depleted. By analyzing several different runs during the plating process you can get a reasonable idea when to add more material to the plating solution.

Analysis of nickel and other cations was done using the UV Cat-1 chemistries. Analysis of all cations and metals could be done in less than 5.2 minutes. In this case nickel was the ion of main interest since its depletion would necessitate the addition of more nickel before it reached a low enough level to cause poor plating. All sample preparation for these runs was done by simply diluting and shooting. Figures 7-9 are of a cation standard and a sample before and after plating.

System suitability can also be used to build control charts for plating solutions at different sampling times during the plating process. Figures 10 and 11 are examples of a sample analyzed before plating (run 1) and at several points during the process (runs 2-4). You can further set up minimum levels allowed and have sample reports flag those runs that fall below the minimum level.

### **EXPERIMENTAL:**

#### **Anion Analysis**

System:	Q4000E
Electrolyte:	Chromate/OFM or Phosphate/OFM
Capillary:	75um X 60cm
Injection:	Hydrostatic, 30 seconds
Run Voltage:	-20kV or -15kV
Detection:	Indirect UV at 254nm or Direct UV at 185nm
Data:	Millennium 2010 Chromatography Manager, Ver. 2.1 with CIA Option

## Cation Analysis

System:	Q4000E
Electrolyte:	UV Cat-1/HIBA
Capillary:	75um X 60cm
Injection:	Hydrostatic, 30 seconds
Run Voltage:	+20kV
Detection:	Indirect UV at 185nm
Data:	Millennium 2010 Chromatography Manager, Ver. 2.10 with CIA Option

## FIGURES:

- Figure 1: Electropherogram of anion/organic acid standard using chromate electrolyte
- Figure 2: Electropherogram of plating solution #58 before plating using chromate electrolyte
- Figure 3: Electropherogram of plating solution #58 after plating using chromate electrolyte
- Figure 4: Electropherogram of anion/organic acid standard using phosphate electrolyte
- Figure 5: Electropherogram of plating solution #53 before plating using phosphate electrolyte
- Figure 6: Electropherogram of plating solution #53 after plating using phosphate electrolyte
- Figure 7: Electropherogram of cation standard using UV Cat-1 electrolyte
- Figure 8: Electropherogram of plating solution #53 before plating using UV Cat-1 electrolyte
- Figure 9: Electropherogram of plating solution #53 after plating using UV Cat-1 electrolyte
- Figure 10: Control chart for organic acids and chloride from a sample before and during the plating process
- Figure 10: Summary Bar chart for organic acids and chloride from a sample before and during the plating process

For Sample: Std2

Vial: 2 Inj: 2 Chan: SATIN

Date Processed 11/30/94 04:14:23 PM

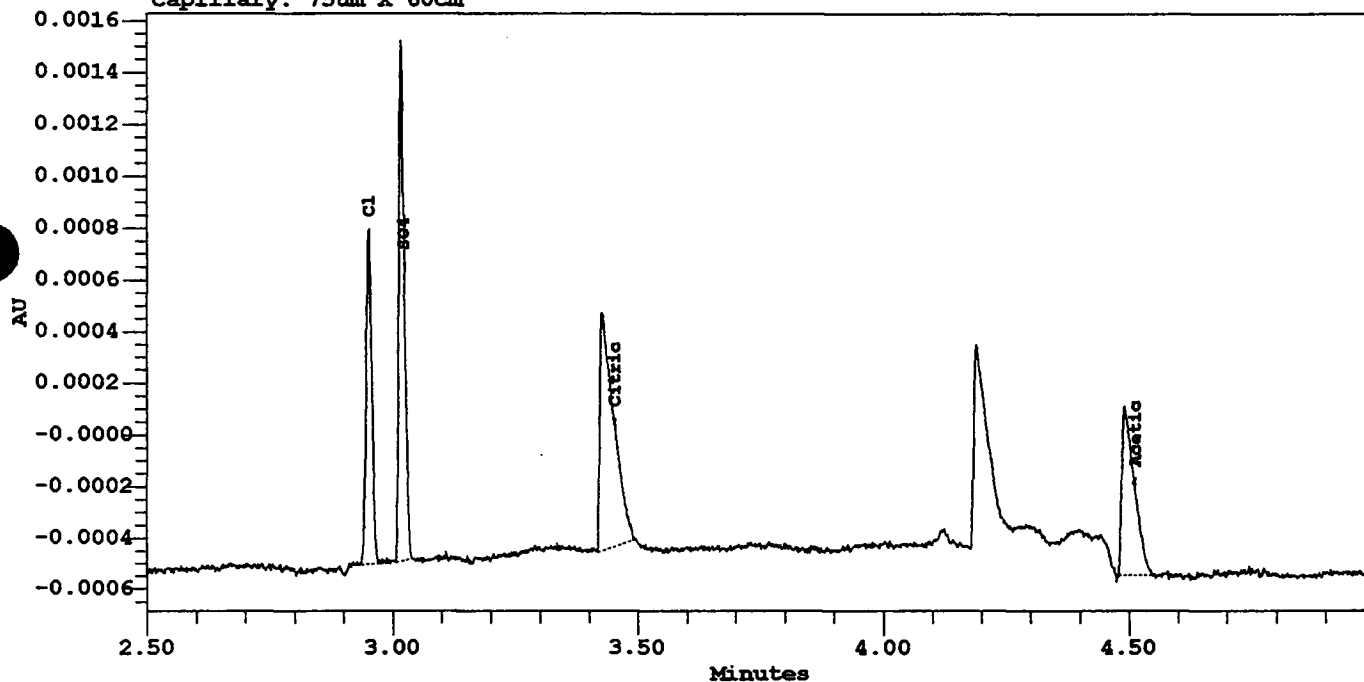
Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA Plating01  
SampleName: Std2  
Vial: 2  
Injection: 2  
Channel: SATIN  
Date Acquired: 11/30/94 03:33:07 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_Anion01  
Processing Method: Plating\_Anions01

Sample Type: Standard  
Volume: 30.00  
Run Time: 5.0 min  
Date Processed: 11/30/94 04:14:23 PM  
Dilution: 1.00000

SampleName: Std2 Electrolyte: 4.0 mM Chromate/0.5 mM OFM RunVoltage: -20kV  
Detection: Indirect 254nm, 0.1 TC InjectionMode: Hydrostatic, 30sec  
Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	2.952	1028	1308	2.000
2	SO4	3.024	1531	2016	4.000
3	Citric	3.455	1826	925	6.000
4	Acetic	4.513	1172	655	2.000

fig. 1

For Sample: 58 Before (0.02 Vial: 4 Inj: 2 Chan: SATIN

Date Processed 11/30/94 04:14:42 PM

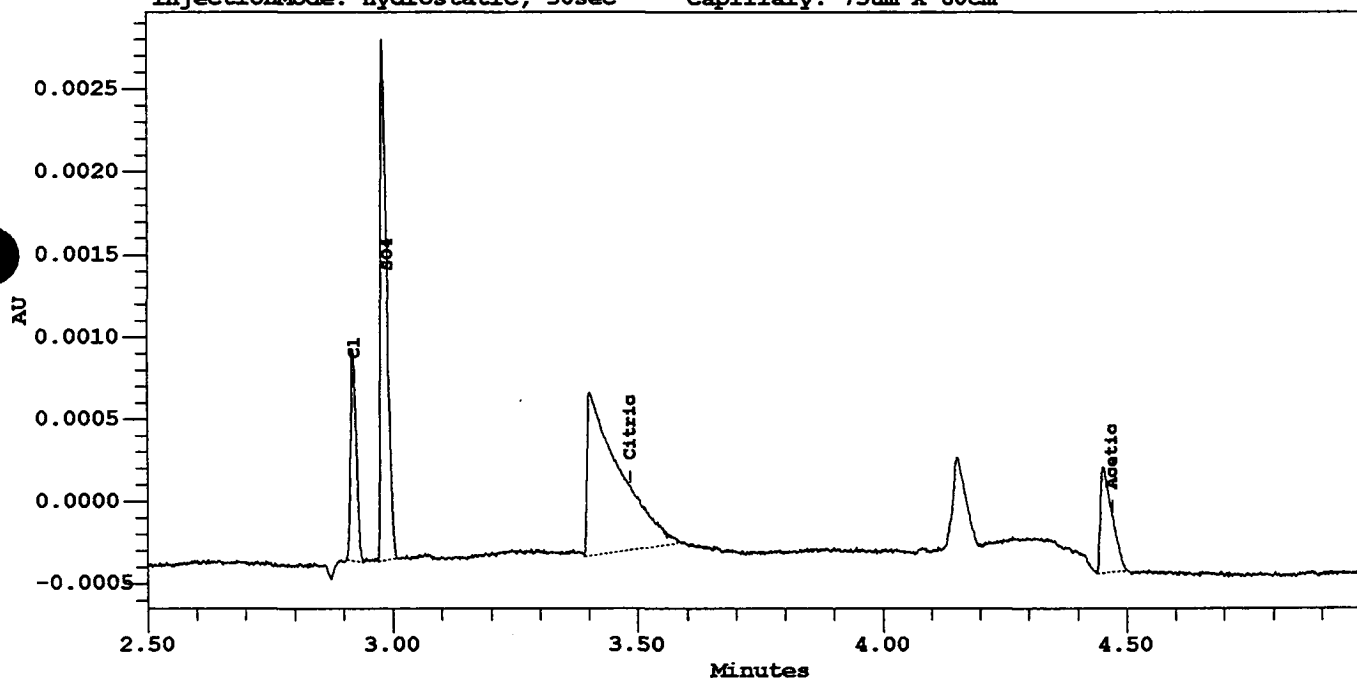
Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA Plating01  
SampleName: 58 Before (0.02-100mL)  
Vial: 4  
Injection: 2  
Channel: SATIN  
Date Acquired: 11/30/94 04:05:42 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA Anion01  
Processing Method: Plating\_Anions01

Sample Type: Unknown  
Volume: 30.00  
Run Time: 5.0 min  
Date Processed: 11/30/94 04:14:42 PM  
Dilution: 5000.00000

SampleName: 58 Before (0.02-100mL) Electrolyte: 4.0 mM Chromate/0.5 mM OFM  
RunVoltage: -20kV Detection: Indirect 254nm, 0.1 TC  
InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



## Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	2.923	1008	1291	9792.928
2	SO4	2.990	2816	3190	36993.650
3	Citric	3.486	4557	990	70253.175
4	Acetic	4.471	1060	651	9342.332

fig. 2

For Sample: 58 After (0.02-Vial: 5 Inj: 2 Chan: SATIN

Date Processed 11/30/94 04:28:09 PM

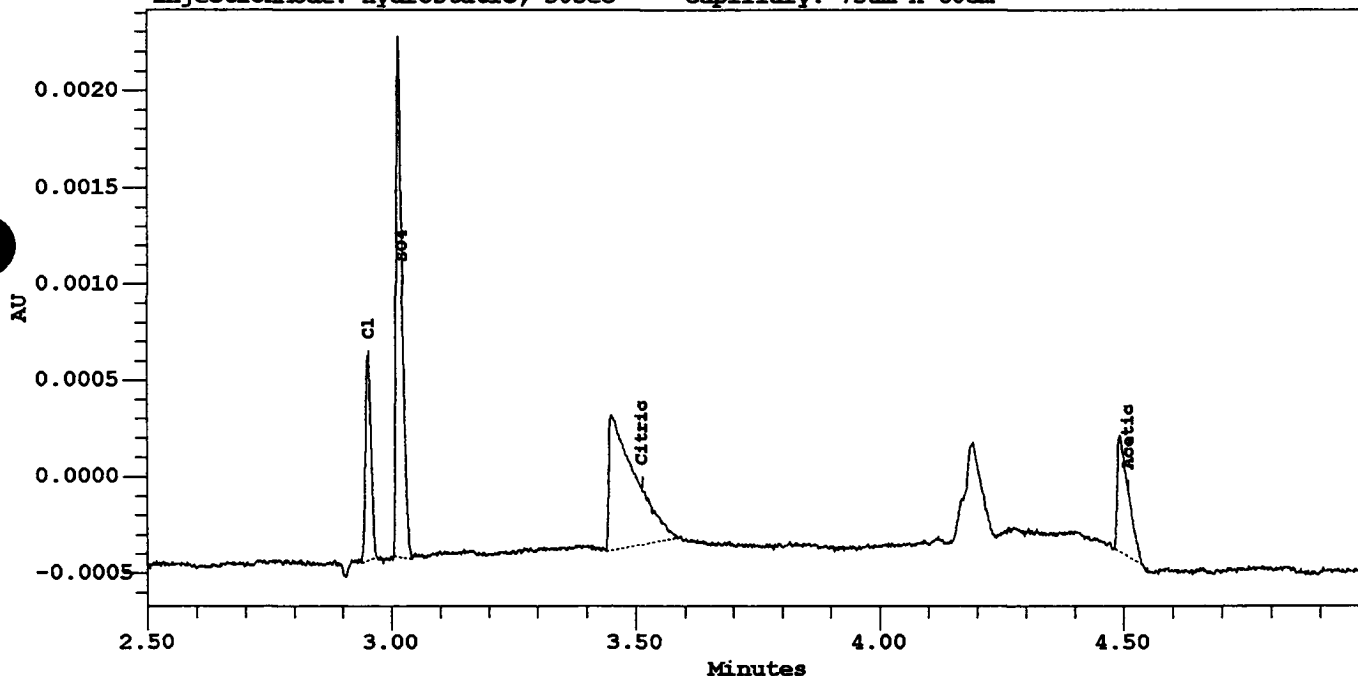
Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA\_Plating01  
SampleName: 58 After (0.02-100mL)  
Vial: 5  
Injection: 2  
Channel: SATIN  
Date Acquired: 11/30/94 04:21:05 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_Anion01  
Processing Method: Plating\_Anions01

Sample Type: Unknown  
Volume: 30.00  
Run Time: 5.0 min  
Date Processed: 11/30/94 04:28:09 PM  
Dilution: 5000.00000

SampleName: 58 After (0.02-100mL) Electrolyte: 4.0 mM Chromate/0.5 mM OFM  
RunVoltage: -20kV Detection: Indirect 254nm, 0.1 TC  
InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	2.955	824	1066	7771.317
2	SO4	3.024	2257	2724	29370.631
3	Citric	3.514	2571	700	40633.495
4	Acetic	4.511	956	610	8361.688

fig. 3

For Sample: Std2

Vial: 3 Inj: 1 Chan: SATIN

Date Processed 12/02/94 12:29:56 PM

Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA\_Plating01

SampleName: Std2

Vial: 3

Injection: 1

Channel: SATIN

Date Acquired: 12/02/94 11:45:44 AM

Scale Factor: 1.00

Acq Meth Set: CIA\_Anion\_5pts

Processing Method: Plating\_Phosphate01

Sample Type: Standard

Volume: 30.00

Run Time: 7.2 min

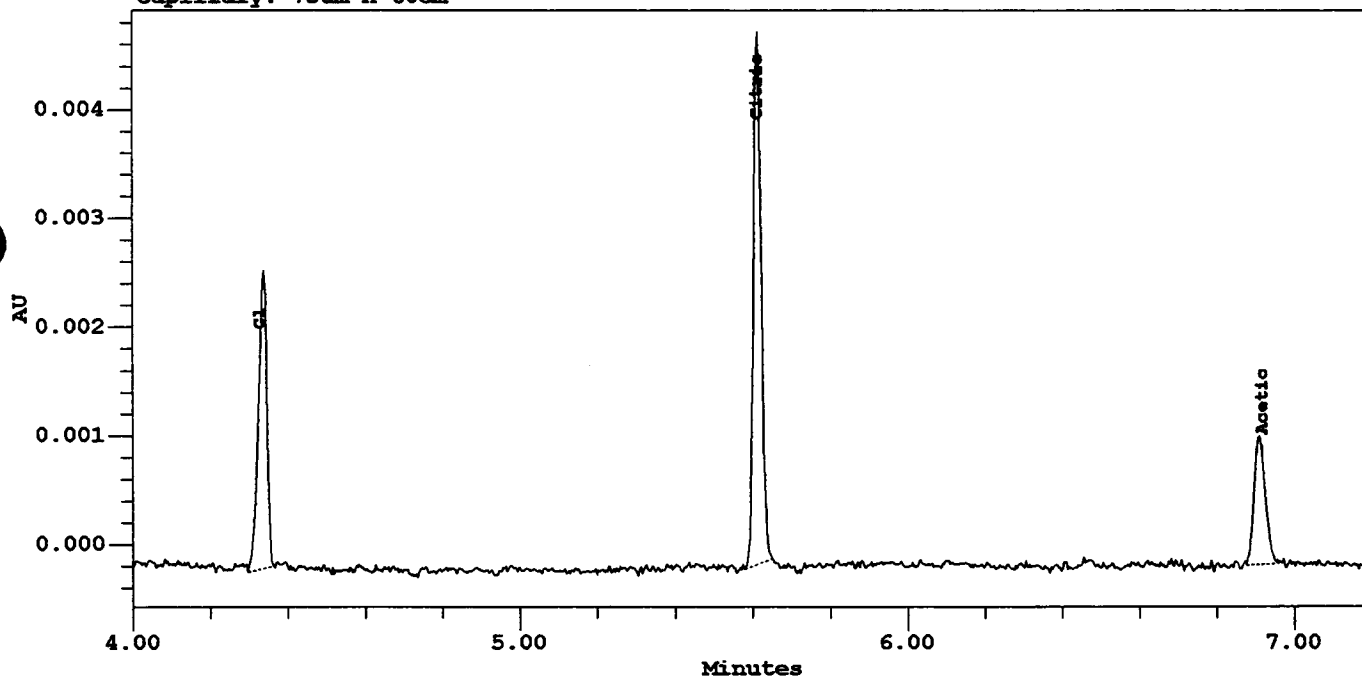
Date Processed: 12/02/94 12:29:56 PM

Dilution: 1.00000

SampleName: Std2 Electrolyte: 25 mM Phosphate/0.5 mM OPM-OH RunVoltage: -15kV

Detection: Direct 185nm, 0.1 TC InjectionMode: Hydrostatic, 30sec

Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	4.330	4125	2737	2.000
2	Citric	5.615	6563	4851	6.000
3	Acetic	6.918	2411	1182	2.000

fig. 4

For Sample: 53 Before (0.02 Vial: 6 Inj: 1 Chan: SATIN

Date Processed 12/02/94 12:38:20 PM

Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

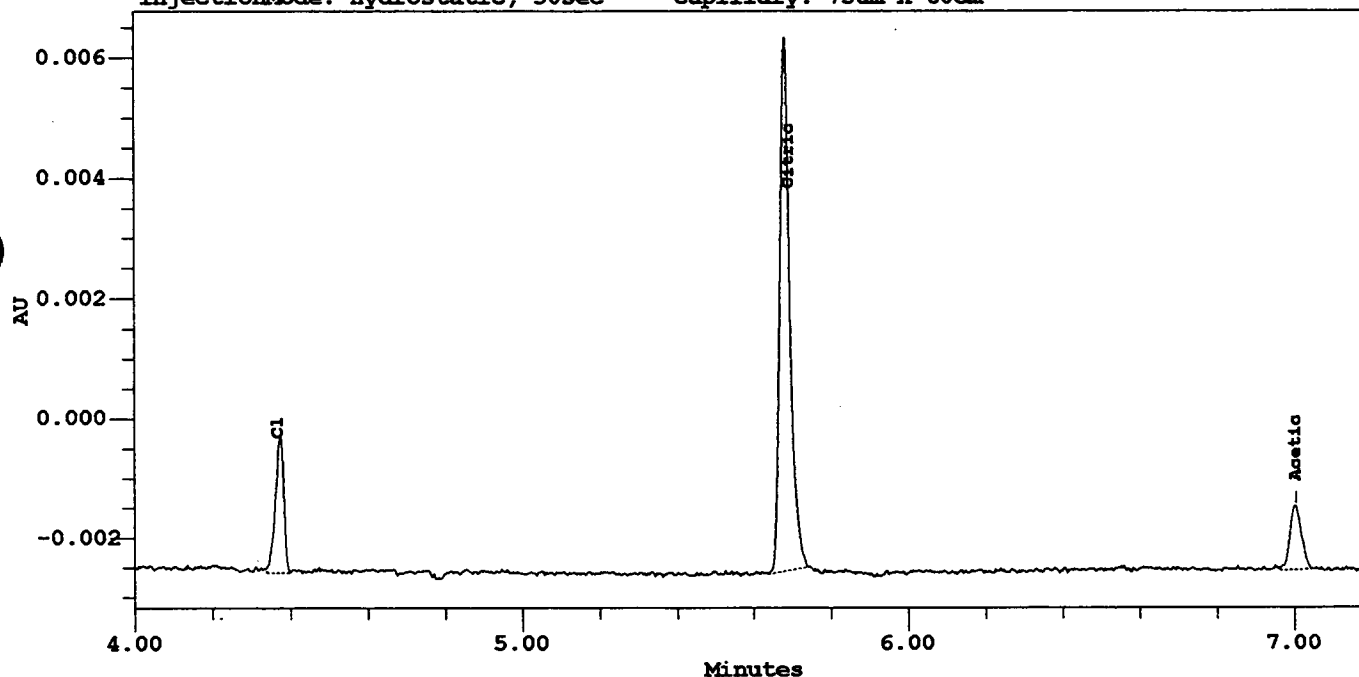
Project Name: CIA\_Plating01  
SampleName: 53 Before (0.02-100mL)  
Vial: 6  
Injection: 1  
Channel: SATIN  
Date Acquired: 12/02/94 12:25:20 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_Anion\_5pts  
Processing Method: Plating\_Phosphate01

Sample Type: Unknown  
Volume: 30.00  
Run Time: 7.2 min  
Date Processed: 12/02/94 12:38:20 PM  
Dilution: 5000.00000

SampleName: 53 Before (0.02-100mL) Electrolyte: 25 mM Phosphate/0.5 mM OFM-OH

RunVoltage: -15kV Detection: Direct 185nm, 0.1 TC

InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



*Peak Results*

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	4.367	3446	2311	8457.296
2	Citric	5.692	16248	8887	73017.185
3	Acetic	7.005	2262	1073	9171.118

fig. 5

For Sample: 53 After (0.02-Vial: 7 Inj: 1 Chan: SATIN

Date Processed 12/02/94 12:57:47 PM

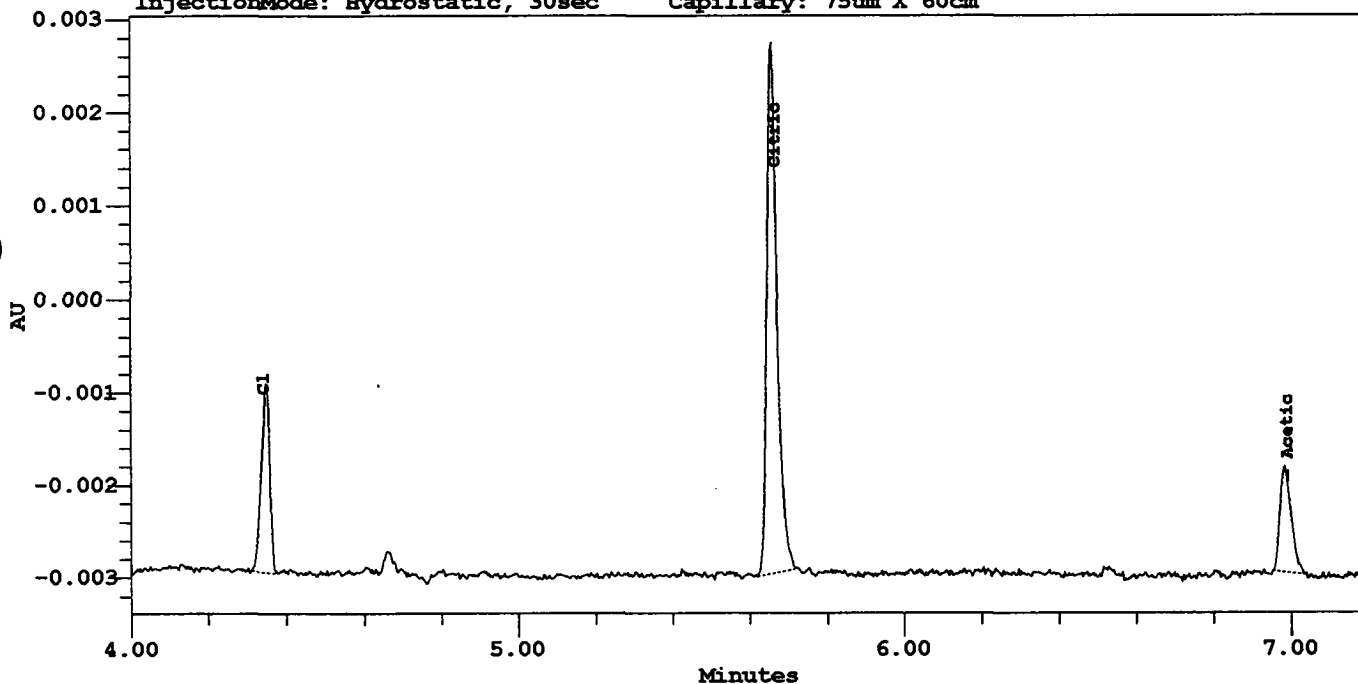
Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA\_Plating01  
SampleName: 53 After (0.02-100mL)  
Vial: 7  
Injection: 1  
Channel: SATIN  
Date Acquired: 12/02/94 12:35:14 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_Anion\_5pts  
Processing Method: Plating\_Phosphate01

Sample Type: Unknown  
Volume: 30.00  
Run Time: 7.2 min  
Date Processed: 12/02/94 12:57:47 PM  
Dilution: 5000.00000

SampleName: 53 After (0.02-100mL) Electrolyte: 25 mM Phosphate/0.5 mM OPM-OH  
RunVoltage: -15kV Detection: Direct 185nm, 0.1 TC  
InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	Cl	4.343	2964	2052	7312.004
2	Citric	5.668	10488	5705	47434.357
3	Acetic	6.993	2309	1149	9366.215

fig. 6



For Sample: Std 3

Vial: 5 Inj: 2 Chan: SATIN

Date Processed 11/30/94 05:41:31 PM

Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA\_Plating01

SampleName: Std 3

Vial: 5

Injection: 2

Channel: SATIN

Date Acquired: 11/28/94 12:56:20 PM

Scale Factor: 1.00

Acq Meth Set: CIA\_01

Processing Method: Plating\_Cations\_02a

Sample Type: Standard

Volume: 30.00

Run Time: 7.0 min

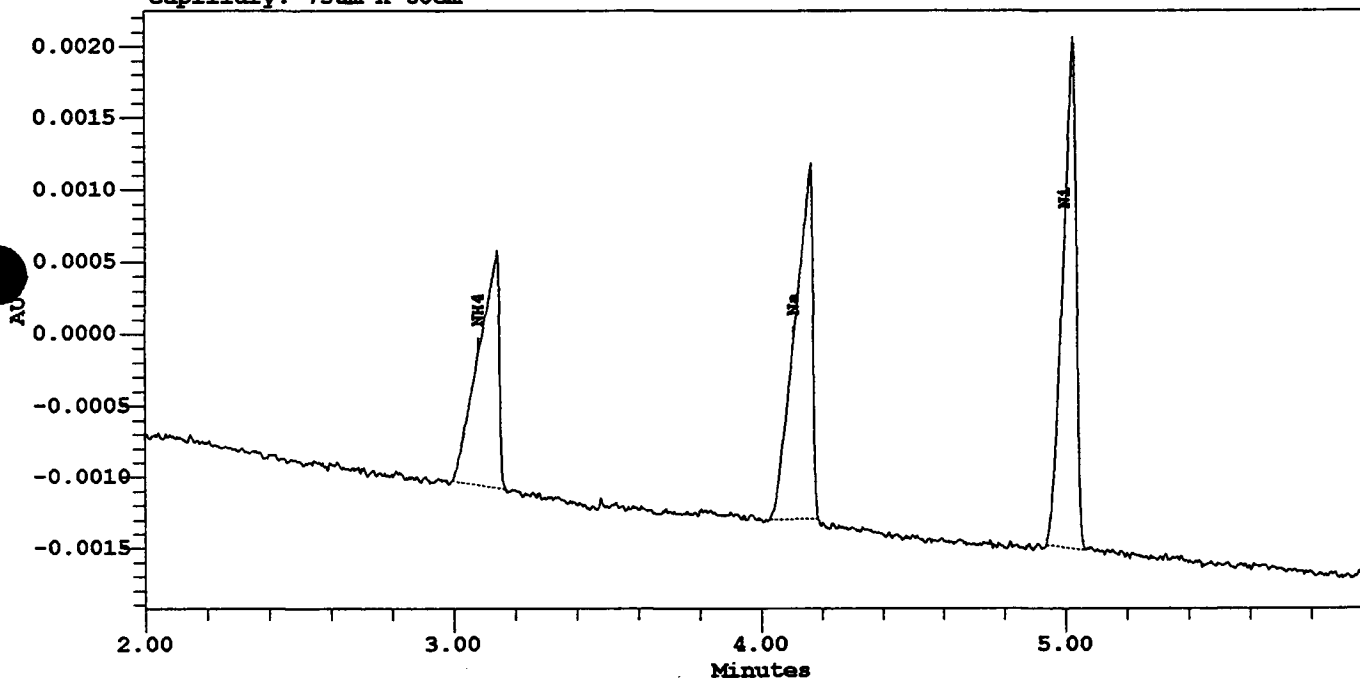
Date Processed: 11/30/94 05:41:31 PM

Dilution: 1.00000

SampleName: Std 3 Electrolyte: 4.0 mM UV Cat-1/6.5 mM HIBA RunVoltage: +20 kV

Detection: Indirect 185nm, 0.3 TC InjectionMode: Hydrostatic, 30sec

Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	NH4	3.082	7728	1657	4.000
2	Na	4.108	10004	2480	4.000
3	Ni	5.000	10901	3570	6.000

fig. 7

For Sample: 53 Before (0.01Vial: 6 Inj: 1 Chan: SATIN

Date Processed 11/30/94 05:42:22 PM

Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

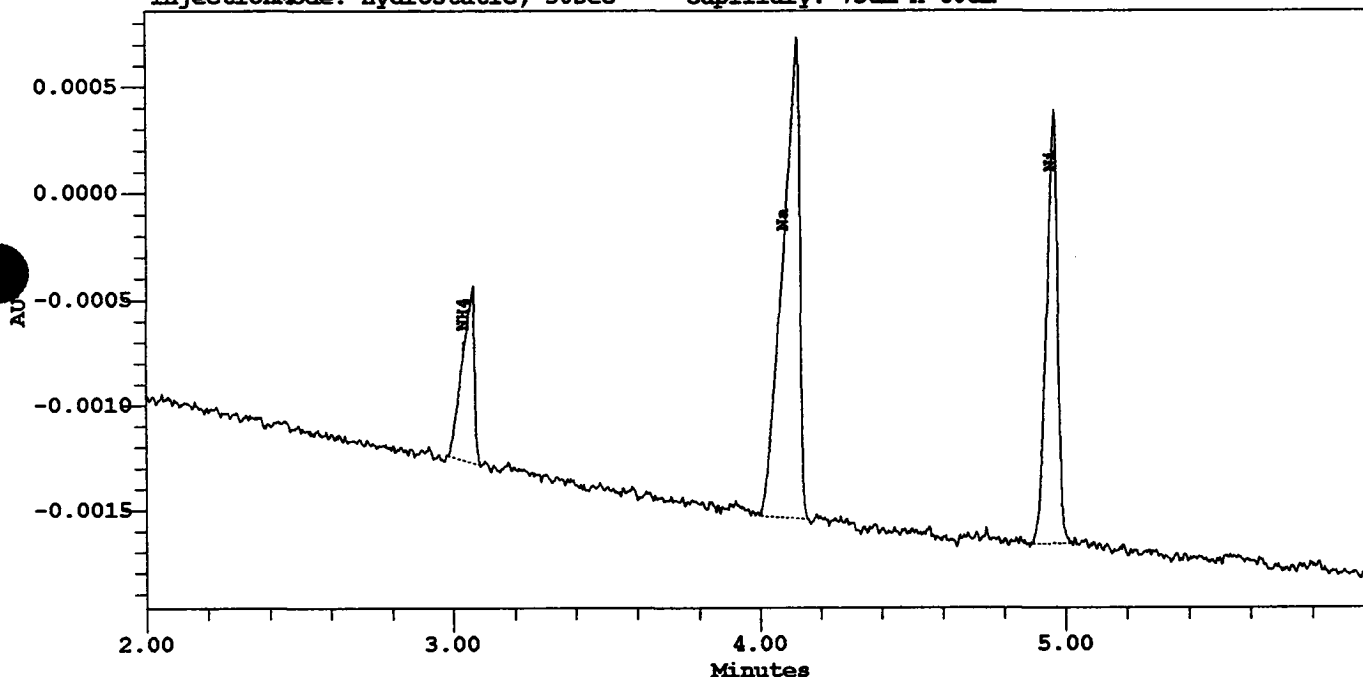
Project Name: CIA\_Plating01  
SampleName: 53 Before (0.01-100mL)  
Vial: 6  
Injection: 1  
Channel: SATIN  
Date Acquired: 11/28/94 01:05:33 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_01  
Processing Method: Plating\_Cations\_02a

Sample Type: Unknown  
Volume: 30.00  
Run Time: 7.0 min  
Date Processed: 11/30/94 05:42:22 PM  
Dilution: 10000.00000

SampleName: 53 Before (0.01-100mL) Electrolyte: 4.0 mM UV Cat-1/6.5 mM HIBA

RunVoltage: +20 kV Detection: Indirect 185nm, 0.3 TC

InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	NH4	3.035	2267	839	12008.021
2	Na	4.080	8715	2273	34760.459
3	Ni	4.953	5227	2052	29785.775

Fig. 8

For Sample: 53 After (0.01-Vial: 8 Inj: 1 Chan: SATIN

Date Processed 11/30/94 05:47:00 PM

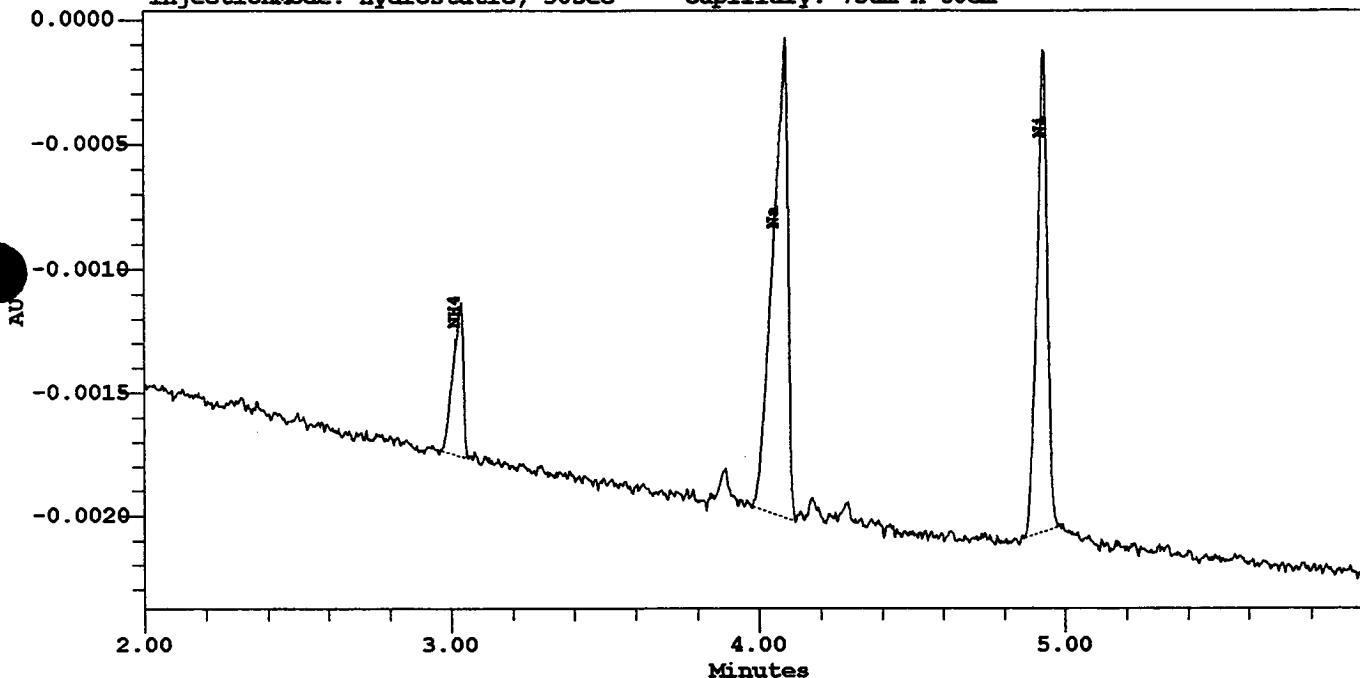
Channel Descr: Indirect UV

Northern Kentucky Laboratory  
Sample Information

Project Name: CIA\_Plating01  
SampleName: 53 After (0.01-100mL)  
Vial: 8  
Injection: 1  
Channel: SATIN  
Date Acquired: 11/28/94 01:42:20 PM  
Scale Factor: 1.00  
Acq Meth Set: CIA\_01  
Processing Method: Plating\_Cations\_02a

Sample Type: Unknown  
Volume: 30.00  
Run Time: 7.0 min  
Date Processed: 11/30/94 05:47:00 PM  
Dilution: 10000.00000

SampleName: 53 After (0.01-100mL) Electrolyte: 4.0 mM UV Cat-1/6.5 mM HIBA  
RunVoltage: +20 kV Detection: Indirect 185nm, 0.3 TC  
InjectionMode: Hydrostatic, 30sec Capillary: 75um X 60cm



## Peak Results

#	Name	Migration Time (min)	Area (uV*sec)	Height (uV)	Amount-ppm
1	NH4	3.013	1402	628	7482.546
2	Na	4.050	6436	1938	25861.046
3	Ni	4.922	4687	1942	26876.732

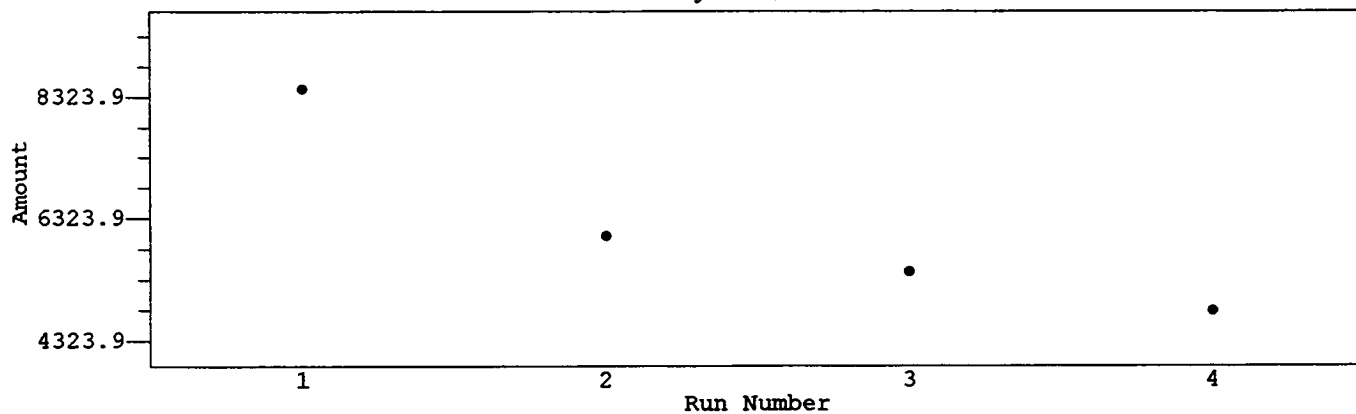
fig. 9

Project Name: CIA\_Plating01

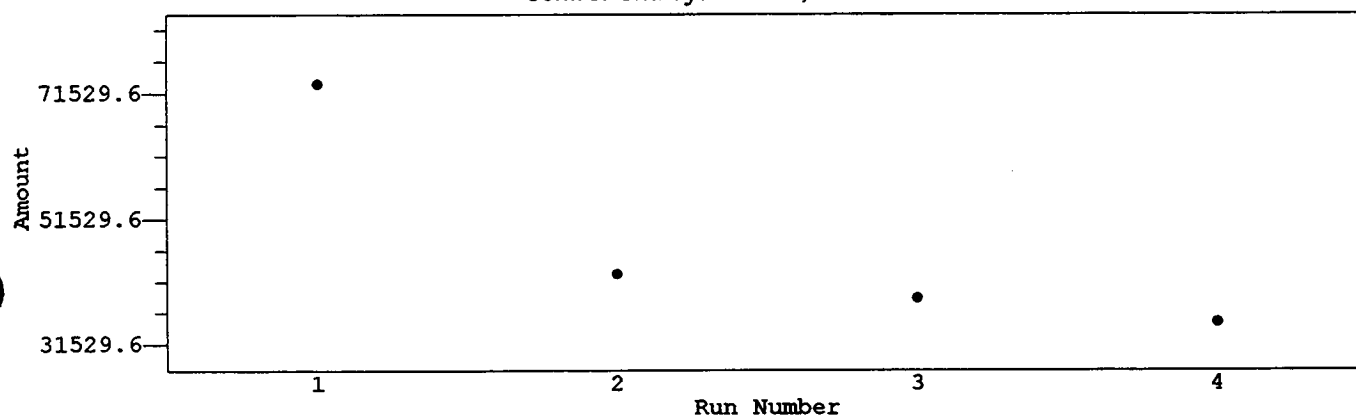
Username: Stuart

Date Prepared: December 2, 1994

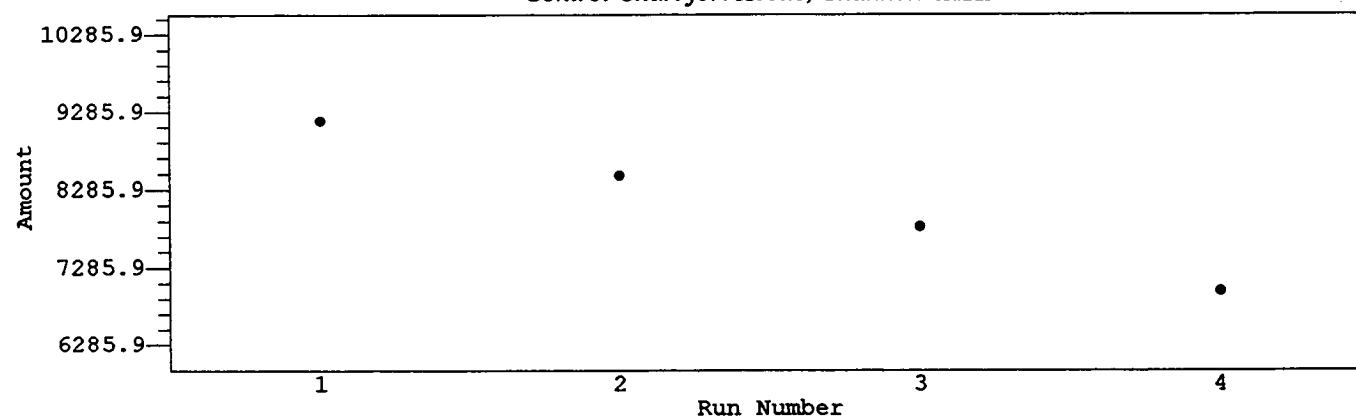
*Control Chart for: Cl, Channel: SATIN*



*Control Chart for: Citric, Channel: SATIN*



*Control Chart for: Acetic, Channel: SATIN*



*Run ID*

#	SampleName	Dilution
1	53 Before (0.02-100mL)	5000.00000
2	53 After#1 (0.02-100mL)	5000.00000
3	53 After#2 (0.02-100mL)	5000.00000
4	53 After#3 (0.02-100mL)	5000.00000

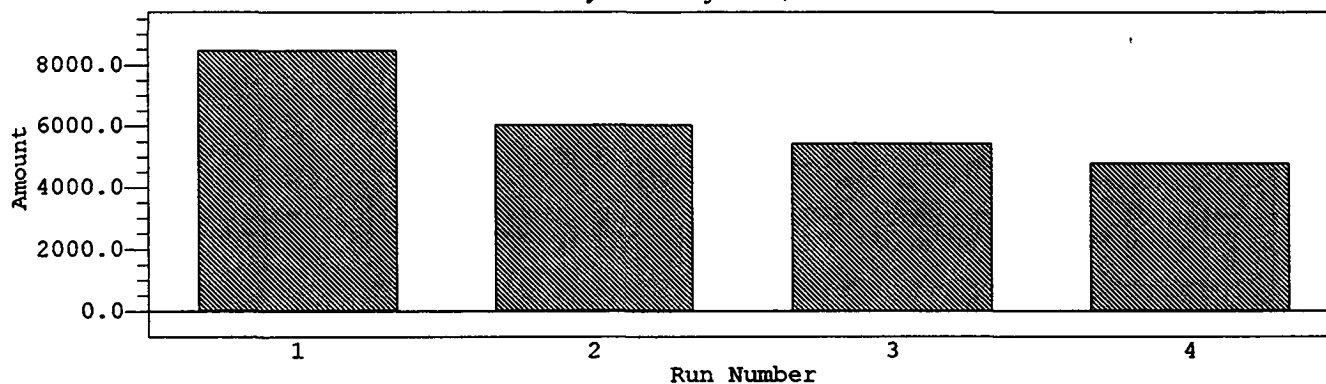
fig. 10

Project Name: CIA\_Plating01

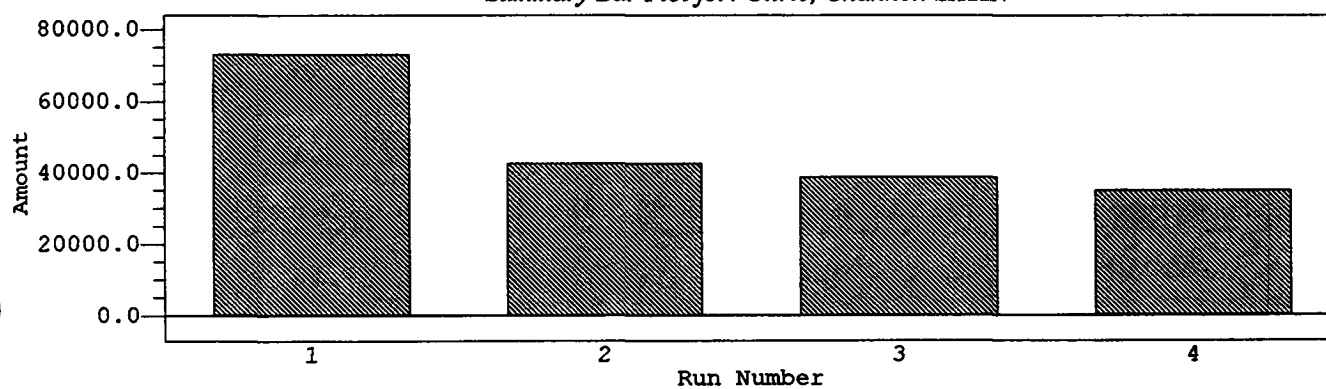
Username: Stuart

Date Prepared: December 2, 1994

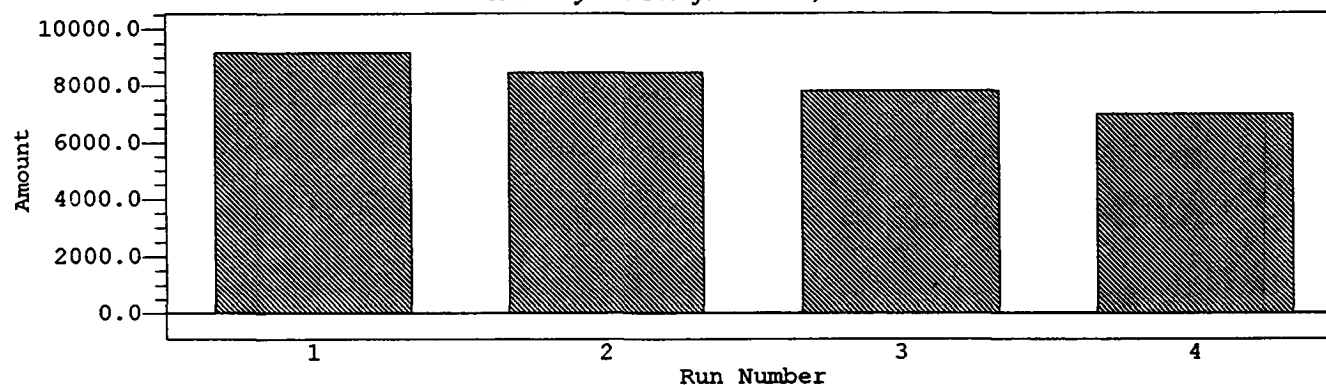
Summary Bar Plot for: Cl, Channel: SATIN



Summary Bar Plot for: Citric, Channel: SATIN



Summary Bar Plot for: Acetic, Channel: SATIN



Run ID

#	SampleName	Dilution
1	53 Before (0.02-100mL)	5000.00000
2	53 After#1 (0.02-100mL)	5000.00000
3	53 After#2 (0.02-100mL)	5000.00000
4	53 After#3 (0.02-100mL)	5000.00000

Fig. 11