

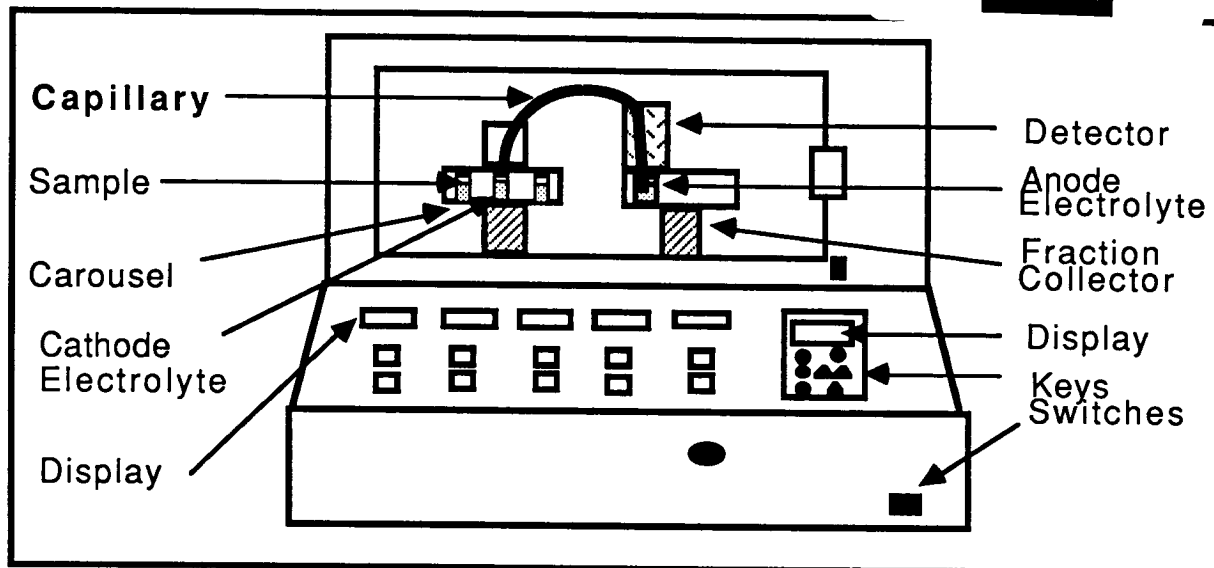
# WATERS ION ANALYSIS METHOD

**Inorganic CE Method - General Anions**  
**High Mobility Electrolyte : Chromate**

Method #  
 N-601

## INSTRUMENTATION

**94-1171**



## CONDITIONS

**Instrument:** Quanta™ 4000

**Electrolyte:** Chromate with OFM™ Anion-BT Chemistry (Patent Applied For)

**Capillary:** Fused Silica (60cm X 75µm)

**Power Supply:** Negative Voltage

**Applied Voltage:** 20 KV

**Current:** 16 µAmp

**Hydrostatic Injection:** 10cm height for 30 seconds

**Detection:** UV at 254nm (Hg lamp)

**Range:** 0.002 AU

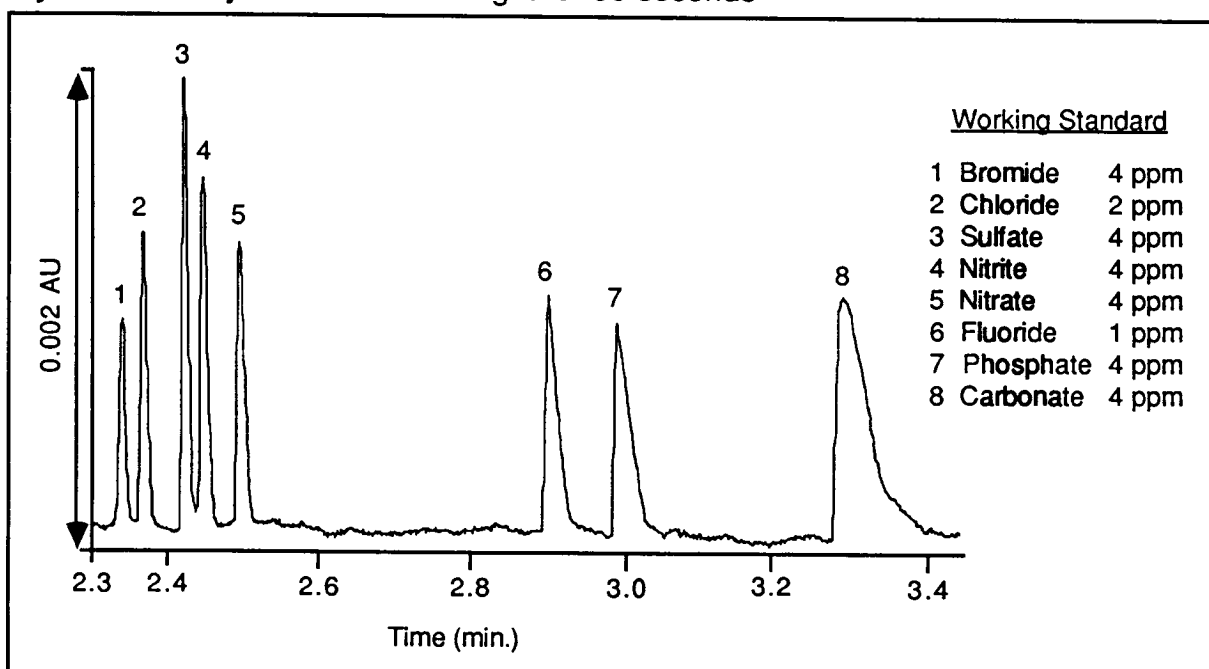
**Detector Polarity:** negative

**Temperature:** Ambient

**Data:** 820 Data System

**Time Constant:** 0.1 seconds

**Sample Rate:** 20 points/second



## ELECTROLYTE PREPARATION

The working electrolyte is prepared by mixing electro-osmotic flow modifier (OFM™ Anion-BT) with electrolyte concentrate.

### I. High Mobility Electrolyte Concentrate (Chromate)

To a 1-liter volumetric flask add:

~500 ml Milli-Q water

23.41g of  $\text{Na}_2\text{CrO}_4$  tetrahydrate (Mallinckrodt AR grade)

68 ml of 10 mN sulfuric acid. [The 10 mN sulfuric acid is prepared by placing 560 microliters of sulfuric acid (Ultrex grade, JT Baker) into a clean 1 liter flask and filling to the mark.]

*100mM Chromate*

Fill the flask to the mark with Milli-Q water and mix thoroughly. This concentrate may be stored in volumetric or sealed glass container for up to 1 year. The 1 liter concentrate makes 20 liters of electrolyte.

### II. Working Electrolyte (Chromate + OFM Anion-BT) (pH 8.0)

To a 200 milliliter volumetric flask add:

5 ml of Waters OFM Anion-BT solution

*= 20mM*

Rinse the pipet or graduated cylinder twice with Milli-Q water, adding the rinses to the volumetric, then add:

10 ml of electrolyte concentrate (chromate)

Fill the flask to the mark with Milli-Q water, mix thoroughly and filter through a  $0.45\mu\text{m}$  Millipore membrane (HA). This results in a chromate electrolyte of pH 8.

**Prepare fresh carrier electrolyte daily.**

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## STANDARD PREPARATION

### I. Concentrated Standard

1000 ppm $\text{F}^-$	0.221 g NaF (ACS) +/- 0.001 g
2000 ppm $\text{Cl}^-$	0.329 g NaCl (ACS) +/- 0.001 g
4000 ppm $\text{NO}_2^-$	0.600 g $\text{NaNO}_2$ (ACS) +/- 0.001 g
4000 ppm $\text{Br}^-$	0.596 g KBr (ACS) +/- 0.001 g
4000 ppm $\text{NO}_3^-$	0.548 g $\text{NaNO}_3$ (ACS) +/- 0.001 g
4000 ppm $\text{HPO}_4^{2-}$	0.569 g $\text{KH}_2\text{PO}_4$ (ACS) +/- 0.001 g
4000 ppm $\text{SO}_4^{2-}$	0.592 g $\text{Na}_2\text{SO}_4$ (ACS) +/- 0.001 g
4000 ppm $\text{HCO}_3^-$	0.707 g $\text{Na}_2\text{CO}_3$ (ACS) +/- 0.001 g

Prepare separate standard concentrates by diluting each of the above to 100 ml with Milli-Q water

## II. Working Standard

Pipet 100  $\mu$ l of each standard concentrate into a 100-ml volumetric flask and dilute with Milli-Q water to result in the working standard (see front page). Prepare fresh working standard weekly. **Always include Bromide and Phosphate in the working standard.**

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### **DETECTION LIMITS:**

2 x Baseline Noise

<u>Analyte</u>	<u>PPB</u>
Br <sup>-</sup>	380.
Cl <sup>-</sup>	145.
SO <sub>4</sub> <sup>2-</sup>	172.
NO <sub>2</sub> <sup>-</sup>	325.
NO <sub>3</sub> <sup>-</sup>	350.
F <sup>-</sup>	89.
HPO <sub>4</sub> <sup>2-</sup>	375.
CO <sub>3</sub> <sup>2-</sup>	360.

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### **COMMENTS:**

1. Always use 4 ml vials filled to the neck with electrolyte.
2. Always use fresh electrolyte with each carousel-load of samples.
3. If fluoride and phosphate co-elute for a given sample, change the electrolyte in the corresponding vial and rerun the sample.
5. For further information or questions, contact your Waters Inorganic Analysis Specialist or the Inorganic Analysis Group in Milford, MA.