

# Transferring HPLC Gradient Methods Using CORTECS Solid-Core Particle Columns

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#### **APPLICATION BENEFITS**

- ~4X decrease in analysis time resulting in faster throughput for routine sample analysis
- ~2X reduction in solvent usage and sample injected
- Operate within the <5,000 psi pressure limits of HPLC instrumentation

#### WATERS SOLUTIONS

Alliance® HPLC System

CORTECS® 2.7 µm Solid-Core
Particle Columns

Empower® 3 Chromatography Data Software

#### **KEY WORDS**

Method transfer, Alliance, HPLC, CORTECS, Abacavir, related substances, solid-core

#### INTRODUCTION

Transferring HPLC gradient methods that use larger volume columns packed with larger particles to smaller volume columns packed with highly-efficient CORTECS 2.7  $\mu$ m Particles is an easy way to reduce analysis time, solvent and sample consumption, and, ultimately, cost. When transferring the HPLC gradient method avoid compromising the chromatographic separation by properly adjusting the method conditions and selecting the equivalent column chemistry.

The following application note demonstrates a proper method transfer of a typical HPLC gradient method for abacavir-related compounds. Abacavir (Ziagen®) is a nucleoside reverse-transcriptase inhibitor that is used in anti-HIV therapy. The sample is composed of five compounds, four of them related substances of the main component abacavir; all shown in Figure 1. A typical column for this type of assay is a fully porous  $C_{18}$ ,  $5~\mu m$ , 4.6~x~150~mm column. The analysis time of this gradient method can be significantly reduced by transferring to a CORTECS  $C_{18}$ ,  $2.7~\mu m$  Column, while maintaining the selectivity and the resolution of the peaks of interest.

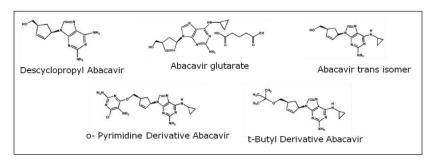


Figure 1. Abacavir and its related substances.

#### **EXPERIMENTAL**

# Method using a fully porous $C_{18}$ , 5 $\mu$ m, 4.6 x 150 mm Column

#### LC conditions

System: Alliance HPLC with

2489 TUV detector

Column: Fully porous C<sub>18</sub>,

5 μm, 4.6 x 150 mm

Mobile phase A: 0.1% trifluoroacetic acid

in water

Mobile phase B: 85% methanol in water

Backpressure: 1,800 psi

Gradient:

Time	Flow rate		
( <u>min</u> )	( <u>mL/min</u> )	<u>%A</u>	<u>%B</u>
Initial	1.0	95	5
23.64	1.0	70	30
38.39	1.0	10	90
43.83	1.0	10	90
44.89	1.0	95	5
50.00	1.0	95	5

Detection: UV at 254 nm

Needle wash: 90:10 water:acetonitrile

Seal wash: 80:20 acetonitrile:water

Injection volume: 8 µL

# Transferred method using a CORTECS $C_{18}$ , 2.7 $\mu$ m, 4.6 x 75 mm Column

#### LC conditions

System: Alliance HPLC with

2489 TUV detector

Column: CORTECS C<sub>18</sub>, 2.7 µm,

4.6 x 75 mm

(p/n 186007376)

Mobile phase A: 0.1% trifluoroacetic acid

in water

Mobile phase B: 85% methanol in water

Backpressure: 4,400 psi

Gradient:

Time	Flow rate		
( <u>min</u> )	(mL/min)	<u>%A</u>	<u>%B</u>
Initial	1.85	95	5
6.38	1.85	70	30
10.37	1.85	10	90
11.83	1.85	10	90
12.12	1.85	95	5
15.00	1.85	95	5

Detection: UV at 254 nm

Needle wash: 90:10 water:acetonitrile
Seal wash: 80:20 acetonitrile:water

Injection volume: 4 µL

Sample vial: Waters LCGC certified clear

glass vial with PTFE/ silicone

septa (p/n 186000307C)

Data Management: Empower 3

#### Sample Preparation

Abacavir-related compounds (USP reference standard) 1.0 mg/mL in 100% HPLC-grade water.

### [APPLICATION NOTE]

#### Method transfer equations

#### Maintaining efficiency when decreasing the particle size

Decreasing the particle size increases the number of theoretical plates in a given column length, therefore, shorter length columns can be used and the separation can be maintained. The following equation is used to determine the appropriate column length when changing particle size.

$$L_{C2} = \frac{L_{C1} \times d_{P2}}{d_{P1}}$$

$$L_{C} = \text{Column Length}$$

$$d_{P} = \text{Particle Size}$$

#### Scaling injection volume

Decreasing column volume requires that the injection volume be adjusted accordingly as described in the following equation.

$$V_{_{I2}} = V_{_{I1}} \left(\frac{d_{_{C2}}}{d_{_{C1}}}\right)^2 \times \left(\frac{L_{_{C2}}}{L_{_{C1}}}\right)$$
  $V_{_{I}} = \text{Injection Volume}$   $V_{_{C}} = \text{Column Length}$   $V_{_{C}} = \text{Column Diameter}$ 

#### Scaling flow rate

Flow rates must be adjusted as column internal diameter changes to maintain the same linear velocity. The flow rates must also be adjusted in inverse proportion to the change in particle size to maintain the performance; this is done using the following equation.

$$F_{\rm C2} = F_{\rm C1} \times \left(\frac{d_{\rm C2}}{d_{\rm C1}}\right)^2 \times \left(\frac{d_{\rm p1}}{d_{\rm p2}}\right) \qquad \begin{array}{l} {\rm F_{\rm C}} = {\rm Flow} \; {\rm Rate} \\ {\rm d_{\rm C}} = {\rm Column} \; {\rm diameter} \\ {\rm d_{\rm P}} = {\rm Particle} \; {\rm Size} \end{array}$$

#### Scaling gradient duration

To maintain the same number of column volumes on both columns, the gradient time must be altered to maintain the gradient slope. The gradient time can be adjusted using the following equation.

$$t_{g2} = t_{g1} \times \left(\frac{F_{c1}}{F_{c2}}\right) \times \left(\frac{d_{c2}}{d_{c1}}\right)^2 \times \left(\frac{L_{c2}}{L_{c1}}\right) \quad \begin{array}{l} F_{c} = Flow \ Rate \\ d_{c} = Column \ diameter \\ d_{P} = Particle \ Size \end{array}$$

#### **RESULTS AND DISCUSSION**

The CORTECS  $C_{18}$ ,  $2.7~\mu m$  chemistry was chosen for the transfer; this was based on the fully porous  $C_{18}$ ,  $5~\mu m$  column typically used for this type of assay. Transferring to a column packed with  $2.7~\mu m$  particles requires a 75~m m column length to maintain the L/dp ratio. For this transfer, a 4.6~x~75~m m column configuration was chosen. The change in column length required that the injection volume be adjusted from  $8~\mu L$  to  $4~\mu L$ . Since the I.D. of the column was not changed the adjusted flow rate is based on the change in particle size only; the adjusted flow rate was calculated to be 1.85~m L/m inute. Adjusting the column configuration and the flow rate requires that each time segment of the gradient also be adjusted to ensure that the separation takes place over the equivalent number of column volumes.

A comparison from the method transfer is shown in Table 1 and in Figure 2. The transferred method using the CORTECS  $C_{18}$ , 2.7  $\mu$ m Column has equivalent selectivity to the method that was performed using a fully porous  $C_{18}$ , 5  $\mu$ m column. Also, the resolution values for two critical peak pairs have been maintained. The 4,400 psi backpressure generated on the CORTECS  $C_{18}$ , 2.7  $\mu$ m, 4.6 x 75 mm Column is well within the 5,000 psi pressure limit of the HPLC instrument. Transferring the HPLC gradient method to the CORTECS  $C_{18}$ , 2.7  $\mu$ m Column reduces the analysis time is by a factor of  $\sim$ 4X and the solvent consumption by  $\sim$ 2X.

Table 1. Results of Abacavir-related substances method transferred ssing a CORTECS  $C_{18}$ , 2.7  $\mu m$  Column.

Column	Column	USP Resolution		Analysis	Volume of	Backpressure
	Dimension	Peaks 3,2	Peaks 4,3	Time (min)	MeCN/Run (mL)	(psi)
Fully Porous C <sub>18</sub> , 5 μm	4.6 x 150 mm	2.7	3.3	50.0	20.8	1800
CORTECS C <sub>18</sub> , 2.7 µm	4.6 x 75 mm	2.7	4.1	15.0	10.5	4400

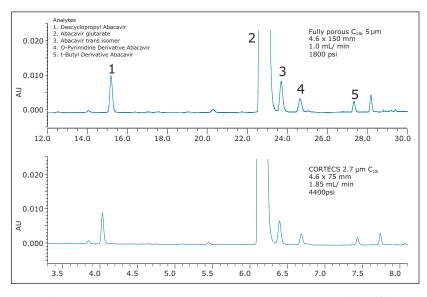


Figure 2. Chromatograms of Abacavir-related substances method transferred to a CORTECS  $C_{18}$ , 2.7  $\mu$ m Column.

### [APPLICATION NOTE]

#### CONCLUSIONS

Transferring HPLC gradient methods that use larger volume columns packed with larger particles to smaller volume columns packed with highly efficient CORTECS particles can easily be achieved. A typical HPLC gradient method for the analysis of abacavir-related substances was successfully transferred to demonstrate a ~4X improvement in throughput while maintaining the chromatographic separation. In addition to the time savings, solvent consumption was reduced by  $\sim$ 2X. The backpressure generated when using the CORTECS  $C_{18}$ , 2.7  $\mu m$ , 4.6 x 75 mm Column is well within the limits of the HPLC instrument of <5,000 psi. When transferring HPLC gradient methods to CORTECS 2.7 µm Columns the increase in throughput and the decrease in solvent consumption add up to significant cost savings.



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