

ANALYSIS OF NERVE AGENT DEGRADATION PRODUCTS BY UPLC/MS/MS

Aisling M. O'Connor, Waters Corporation, Milford MA 01757

INTRODUCTION

The need for a rapid sensitive method for the detection and identification of the presence of Chemical Warfare Agents has been highlighted in recent years due to increased public concern about the use of chemical warfare. Although, more than 160 countries have signed the Chemical Weapons Convention (CWC) which prohibits the development, production, stockpiling and use of chemical weapons by the military^[1] there is a serious threat of their use in a terrorist attack.

Nerve agents are organophosphate type compounds. The first class of nerve agents, the G-Series were synthesized in Germany during World War II. The G-series consists of GA (Tabun), GB (Sarin), GD (Soman) and GF (Cyclosarin)^[2]. Sarin gained notoriety in 1995 due to its use in the Tokyo underground terrorist attack^[3]. The second, more toxic series of nerve agents, the V-series consists of VX and R-VX (Russian VX).

All nerve agents are extremely toxic even in small amounts. 1 mg of VX on the skin can be fatal. Their mode of action in the body is to inhibit cholinesterase enzymes causing the accumulation of acetylcholine, resulting in prolonged stimulation and paralysis of muscles^[4].

In the body and in the environment, nerve agents metabolize / degrade rapidly to form their corresponding Alkyl Methyl- Phosphonic Acid (AMPA). Figure 1 shows the nerve agents and their degradation products. A fast, easy analysis method for the AMPAs of the nerve agents, VX, GB, R-VX, GF and GD is presented here. This method

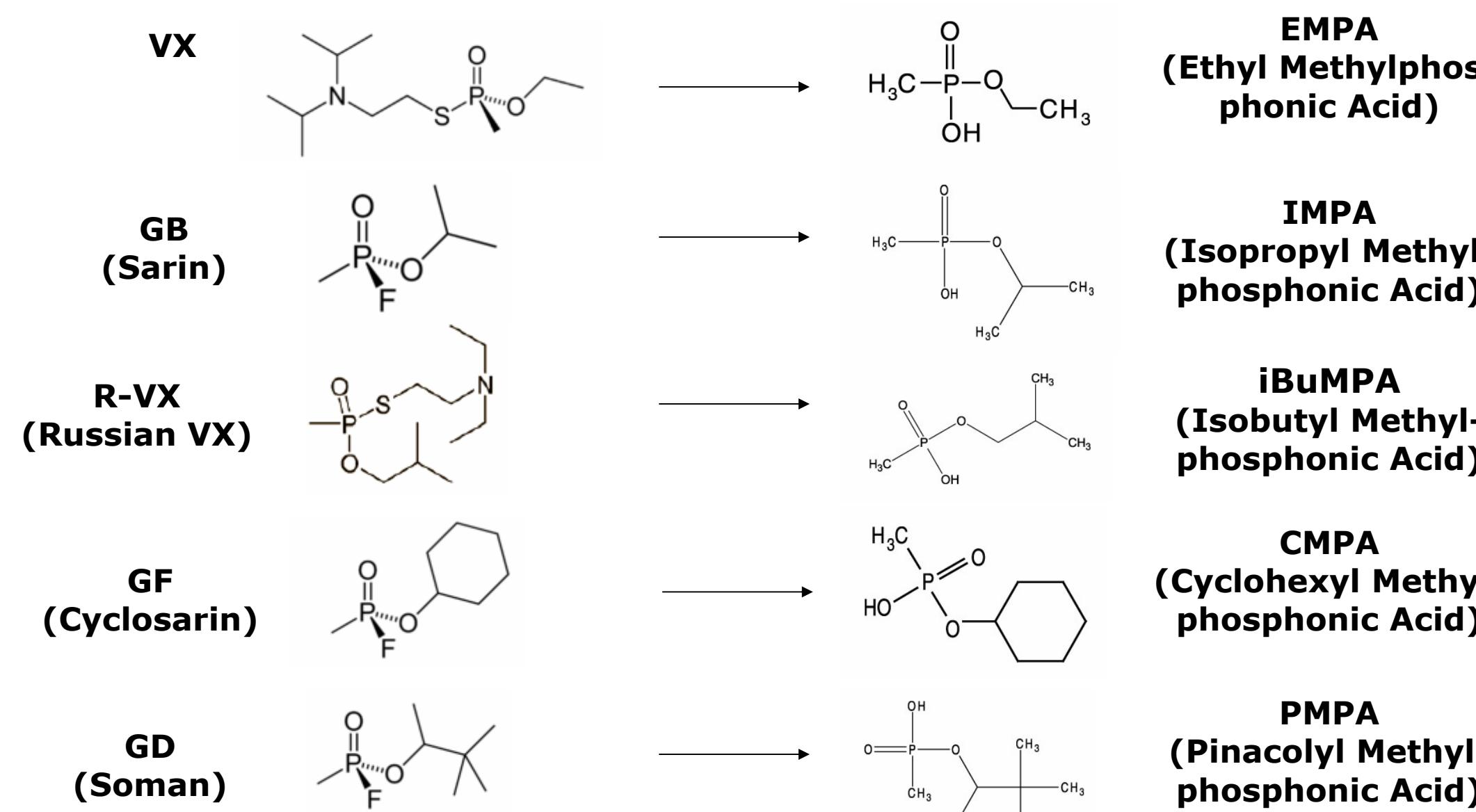


Figure 1: Nerve agents and corresponding degradation products (AMPAs)

is suitable for the analysis of biological and environmental samples, following appropriate sample preparation without the need for derivatization. Five nerve agent degradation products were analyzed using UPLC/MS/MS with the ACQUITY® UPLC TQD system in less than 1.8 minutes; the total cycle time was 4 minutes.



Figure 2: Waters ACQUITY UPLC / TQD System

EXPERIMENTAL CONDITIONS

UPLC Conditions

Instrument: Waters ACQUITY UPLC
Column: Waters ACQUITY UPLC HILIC 2.1x100 mm, 1.7μm
Eluent A: 10 mM Ammonium Acetate in 90 / 10 Acetonitrile / Water
Eluent B: 10 mM Ammonium Acetate in 10 / 90 Acetonitrile / Water
Gradient:

Time (min)	Flow (mL/min)	A (%)	B (%)	Curve
0	0.7	97	3	6
2	0.7	97	3	6
3	0.7	60	40	6
3.1	0.7	97	3	1

Run Time: 4 minutes
Column Temperature: 35°C
Injection volume: 10 μL

MS Conditions

Instrument: Waters ACQUITY TQD
Ionization Mode: ES-
Capillary Voltage: 3.6 kV
Source Temp: 150 °C
Desolvation Temp: 400° C
Desolvation Gas: 800 L/hr

Metabolite	MRM Transition	CV (V)	CE (V)
EMPA	123.1 > 95 123.1 > 79	28 28	12 20
IMPA	137.1 > 95 137.1 > 79	30 30	14 21
iBuMPA	151.1 > 95 151.1 > 77	37 37	13 20
CMPA	177.1 > 95 177.1 > 79	35 35	16 31
PMPA	179.1 > 95 179.1 > 79	35 35	16 32

RESULTS & DISCUSSION

Ultra Performance Liquid Chromatography (UPLC) enables faster and higher peak capacity separations than conventional HPLC. Using UPLC, baseline separation of 5 AMPAs was achieved in under 1.8 minutes compared with up to 20 minute literature reports of HPLC separations [6, 7]. Figure 3 shows the UPLC/MS/MS chromatogram of a 10 pg/μL standard.

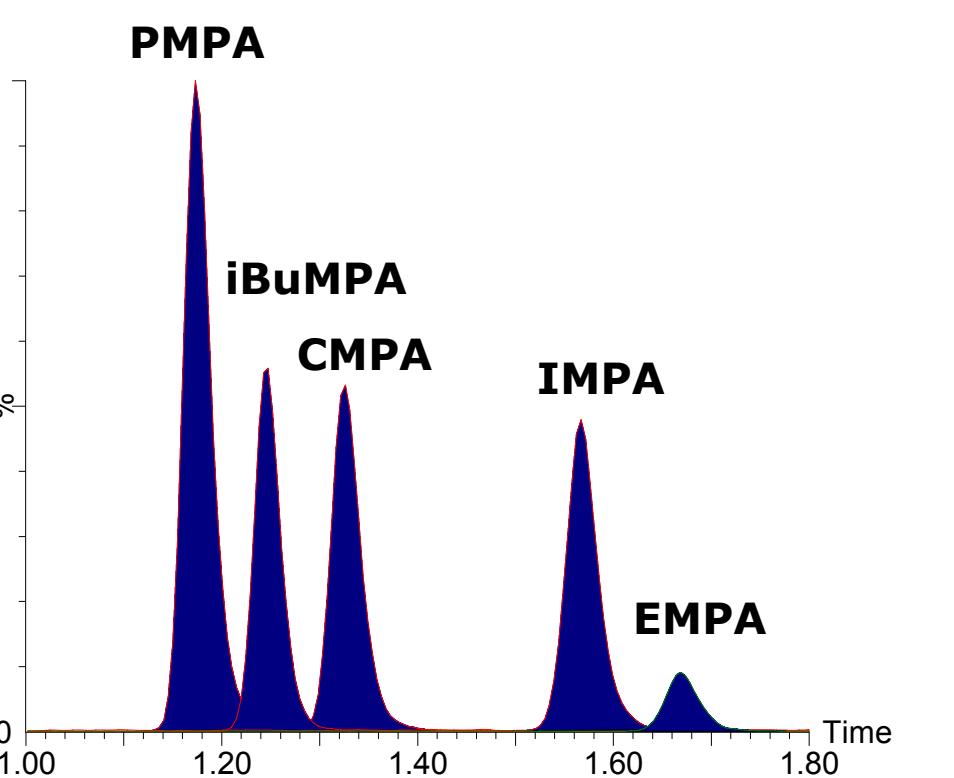


Figure 3: UPLC/MS/MS chromatogram of a standard containing five AMPAs

Compound name: IMPA
Correlation coefficient: r = 0.999454, r² = 0.998907
Calibration curve: 87.64 * x + 24.1722
Response type: External Std, Area
Curve type: Linear, Origin: Exclude, Weighting: 1/x, Axis trans: None

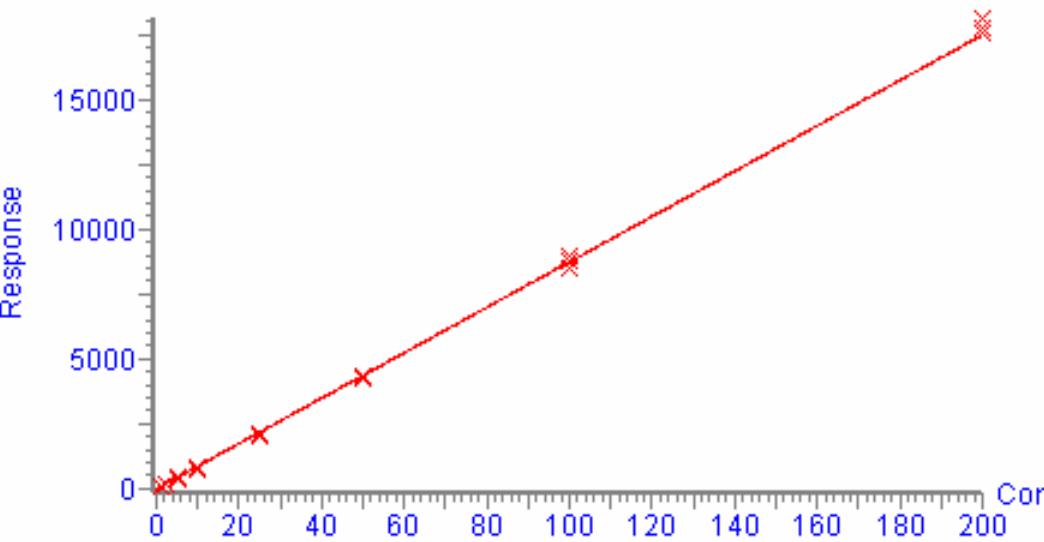


Figure 4: Calibration curve for IMPA

CONCLUSIONS

Quantification and confirmation of the presence of the nerve agents VX, GB, R-VX, GF and GD is achieved in one analysis. This was accomplished by applying a fast, sensitive UPLC/MS/MS method for the analysis of 5 nerve agent degradation products. Separation of the 5 compounds was achieved in less than 1.8 minutes. The method is based on hydrophilic interaction chromatography. This eliminates the time consuming evaporation and reconstitution steps following SPE or protein precipitation.

REFERENCES

- S. Le Moulec, L. Truong, C. Montauban, A. Begos, V. Pichon, B. Bellier, J. Chromatogr. A 1139 (2007) 171-177
- http://en.wikipedia.org/wiki/Nerve_agent
- <http://www.cfr.org/publication/9238/#3>
- New York State Department of Health, "The Facts About Nerve Agents", Fact Sheet, 2004
- E.S. Grumbach, D.M. Diehl, B. Alden, P. Iraneta, U. Neue, J. Mazzeo, Waters Corporation Poster Reprint WA20794, 2003
- P.A. D'Agostino, J.R. Hancock, C.L. Chenier, C.R. Jackson Lepage, J. Chromatogr. A 1110 (2006) 86-94
- Q. Lin, X. Hiu, J. Xie, Analytica Chimica Acta 512 (2004) 93–101