FLAVONOID GLYCOCONJUGATES PROFILING IN LEAVES OF BLUE LUPINE (Lupinus angustifolius) USING LC/MSⁿ AND UPLC/MS SYSTEMS

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INTRODUCTION

Flavonoids and isoflavonoids are an important class of plant phenolic secondary metabolites. They play an important role in various biochemical and physiological processes connected with plant response to biotic and abiotic stresses.



Flavonoids are usually accumulated in tissues as glycosides, frequently acylated with organic acids (e.g. malonic acid), with a large number of isomeric conjugates differing in the hydroxylation pattern of the aglycone as well as position and pattern of glycosilation and acylation. Identification of different derivatives may be achieved using tandem LC/MS techniques. In addition, tandem MS can be used to distinguish between O- and C- glycosylated flavonoids. Fragmentation induced by CID gives rise to Y_0 and B_0 ions for O-glycosides but result in cross ring cleavages (X ions) for C-glycosides (Fig. 1)

We report here the results of profiling of flavonoid conjugates in leaves of blue lupine (*Lupinus angustifolius*). Phenolic secondary metabolites were extracted from tissues with 80% methanol, SPE was performed on RP C18 columns. Samples (5 µL) were analyzed using LC (RP C18 column)-IT MS (Esquire 3000, Bruker Daltonics, Germany)

UPLC (RP C18 column)-TOF MS (LCT Premier, Waters, UK) analysis was performed on the extracts $(1 \ \mu L)$ to investigate the potential of UPLC-MS technology for natural product investigations. Data were acquired simultaneously at low and high source voltages (aperture 1 voltage) to generate both molecular weight and fragment information in a single analysis.

CONCLUSIONS

- •Using traditional LC-IT MSⁿ analysis of the plant extract 26 different flavonoid and isoflavonoid conjugates (Table 1) were separated and identified in about 45 min. Typical single ion chromatograms are shown in Fig. 2.
- •Isomeric conjugates with the same molecular mass but different aglycones and/or glycosilation pattern were distinguished (Fig. 3).
- •The use of UPLC instead of HPLC provides increased resolution and much shorter analysis times; UPLC separation performed in just 15 min! Typical single ion chromatograms are shown in Fig. 4
- •Preliminary UPLC-Tof pseudo MS/MS experiments (using aperture 1 switching) were used to generate structurally diagnostic fragmentation data (Fig. 5b).
- •Exact mass measurement and i-FIT can be used to identify the correct chemical formula of flavonoid and isoflavonoid conjugates (Fig. 6)

RESULTS





Figure 3: Example fragmentation data obtained on LC-MSⁿ (Esquire 3000 Ion Trap) analysis (a) genistein 4',7-O-diglucoside dimalonylated (b) methylluteolin xylosylglucoside dimalonylated

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DISCUSSION

A profile of the flavonoid conjugates extracted from leaves of blue lupine (*Lupinus angustifolius*) using LC-MSⁿ has been performed. LC-MSⁿ technology enabled identification of 26 flavonoid glucoconjugates (Table 1).

Both isofalvone and flavone conjugates were identified. CID MSⁿ spectra permitted us to some extent to propose the glycosilation patterns (Fig 3,5a), it was possible to evaluate the degree of acylation of flavonoid conjugates with malonic acid. However, MSⁿ spectra registered for aglycone ions in the mass spectrum of minor compounds did not give structural information about flavone or isoflavone.

We have demonstrated that the UPLC can drastically reduce LC analysis times. The enhanced chromatographic resolution has in many cases enabled us to identify analytes that may be additional structural isomers (Table 1). For example, two flavonoid conjugates with m/z 767 have been identified and assigned using the data obtained on LC-MSⁿ analysis. However, the improved resolution obtained using UPLC (Fig. 4) enables us to identify an additional peak that may well be caused by another structural isomer.

Our preliminary studies to obtain fragmentation data us-

C27H21O15	t_{r} (min)	3 089	6 56	6 75		
595 1663	measured m/z	595 1642	595 1653	595 1665		
	error (ppm)	3.5	1.7	1.3		
	i-FIT score	1.4	6	9.6		
C ₃₀ H ₃₃ O ₁₈	t _r (min)	4.45	4.65	7.36	7.57	7.84
681.1667	measured <i>m/z</i>	681.1677	681.1677	681.172	681.1665	681.1671
	error (ppm)	1.5	1.3	7.8	0.3	0.1
	i-FIT score	1	0.5	2	2.8	24
C ₃₃ H ₄₁ O ₂₀	t _r (min)	4.96	5.15			
757.2191	measured <i>m/z</i>	757.2180	757.2175			
	error (ppm)	1.5	2.1			
	i-FIT score	0.1	0			
$C_{21}H_{21}O_{11}$	t _r (min)	5.39				
449.1085	measured <i>m/z</i>	449.1073				
	error (ppm)	2.4				
	i-FIT score	0.2				
$C_{27}H_{31}O_{16}$	t _r (min)	5				
611.1612	measured <i>m/z</i>	611.1614				
	error (ppm)	0.3				
	i-FIT score	1				
$C_{36}H_{43}O_{23}$	t _r (min)	6.1	5.93	5.5	5.67	
843.2195	measured <i>m/z</i>	843.2200	843.2181	843.2164	843.2161	
	error (ppm)	0.6	1.7	3.7	4	
	i-FIT score	1.1	0.9	0.1	0.9	
$C_{33}H_{35}O_{21}$	t _r (min)	5.74	8.82	8.11		
767.1671	measured <i>m/z</i>	767.1664	767.1635	767.1682		
	error (ppm)	0.9	4.7	1.4		
	i-FIT score	5.74	8.82	8.11		

Table 2:Summary of ions detected with correct chemical
formulae for selected flavonoid and isoflavonoids in the Lu
pinus angustifolius extract (obtained on UPLC-MS (LCT Tof)
analysis (single scan data)





ing aperture 1 switching on the LCT have been successful. Fragmentation data was obtained for most major peaks (example data Fig. 5B and 6B). Potential compounds of interest could be screened by plotting exact mass chromatograms of diagnostic fragment ions from the high energy data. This technique however is limited if co-elution of two analytes occurs, particularly when common fragment ions are produced. Though the information generated provides a sound basis for deciding which targeted UPLC/MS/MS analysis to perform.

Using exact mass data (in most cases < 3 ppm) and the i-FIT software (Fig. 6) we have been able to propose the chemical formulae of many of the ions giving rise to peaks in the selected ion chromatograms.

Figure 6: (a) Illustration of elemental composition analysis calculator with mass measurement errors and determined elemental composition for methylluteolin xylosylglucoside malonylated (b) psuedo MS/MS data

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