LIPIDOMICS: STUDY OF TOTAL PHOSPHOLIPIDS IN IMMORTALIZED LIVER CELLS EXPOSED TO DIFFERENT FATTY ACID SUBSTRATES

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ABSTRACT

Accumulation of fatty acids into non-adipose tissues can lead to cell dysfunction and cell death. Lipotoxicity from accumulation of long chain fatty acids is specific to, or made more severe by saturated fatty acids. Using H4IIE rat hepatoma cells as a model system of fatty liver disease, changes to the cellular phospholipid pool following exposure to saturated or unsaturated fatty acids were studied.

INTRODUCTION

Metabolic syndrome is a collection of diseases including hyperlipidaemia, obesity, and diabetes mellitus. These diseases share a number of interdependent components: insulin resistance, impaired fatty acid metabolism, and altered lipid profiles. In the liver, these metabolic processes result in a progression of diseases known collectively as nonalcoholic fatty liver disease.¹ Nonalcoholic fatty liver disease (NAFLD) is a chronic disease syndrome initially characterized by the accumulation of fatty acids in the liver (steatosis) with progression, in some individuals, to nonalcoholic steatohepatitis (NASH) and end-stage liver disease

It has been proposed that the trigger for progression into more advanced stages of NAFLD involves an acute insult, or "second hit", superimposed on hepatic steatosis.² The identity of this acute insult, including the cellular intermediates, and signaling pathways used to transmit its signal have not been established. We have recently demonstrated that hepatic steatosis characterized by increased saturated fatty acids leads to increased caspase-3 activity, endoplasmic reticulum stress and liver injury in rats.³ In addition, we have demonstrated that the saturated fatty acids, stearate and palmitate, potently induce endoplasmic reticulum stress and apoptosis in H4IIE liver cells, independently of ceramide accumulation.⁴ Thus, the presence of increased saturated fatty acid delivery to or accumulation in the liver may constitute an intrinsic second hit in the steatotic liver. The present study sought to characterize the cellular lipid environment in H4IIE liver cells following incubation with either palmitate or oleate.

A number of analytical techniques have been used to study the lipid fraction of cells and biological membranes. In particular, analysis of the non volatile membrane components such as phospholipids has relied on ESI-MS. Gross and other researchers have used infusion and nanoelectrospray based analytical strategies based on lipid class prefractionation and tandem mass spectrometry techniques with triple guadrupole mass spectrometers.⁵⁻⁸ Although rich in information content these strategies do not allow for the discovery of new molecular species.

Separation of these lipids is complicated by the variable phospholipid head group and the distribution of fatty acyl chain length and degree of unsaturation. Most HPLC separations of lipids utilize normal phase separations that are plagued by poor reproducibility and poor peak capacity. For this reason, a reversed phase HPLC method using water/ acetonitrile and a C₈ column packed with highly efficient sub 2 µm particles coupled with ESI TOF mass spectrometry was developed. In addition to the accurate mass measurements that are used to indicate phospholipid class, fatty acyl chain total carbon number, and degree of unsaturation for each component; the novel MS^E technique was used to determine the identity of the individual fatty acids.⁹ MS^E delivers parallel precursor and product ion analysis with exact mass in a single run. This is particularly useful in UPLC separations where qualitative and quantitative data is needed.



Major Phospholipid Classes



Phosphatidylinositol

METHODS

Cell Culture and Sample Preparation

H4IIE cells were cultured in DMEM and 10% FBS. At 90% confluence, the media was supplemented with either 500 µM palmitic or oleic acid, bound to albumin, for 6 hours. A separate culture was used as a control Total cellular lipids were extracted using chloroform:methanol and diluted 3:1 prior to analysis.

Liquid Chromatography/Mass Spectrometry

Waters ACQUITY UPLC

Mobile Phases 50 mM Aqueous Ammonium Acetate pH 5.0 Acetonitrile Column ACQUITY BEH C8 2.1x100 mm 1.7 µm d_n Flow Rate 550 µL/min Temperature 50 ° C Gradient Hold 35% B 0.1 minute Ramp to 65% B in 1 minute

Ramp to 95% B in 13 minutes Hold 2 minutes

Waters Q-Tof Premier ESI Negative Mode Capillary= 3.5 kV Cone Voltage= 40 V Collision Energy = 4 eVLockSpray using Leucine Enkephalin

W Mode resolution (fwhm) = 15,000 @ m/z 554

MS^E Experiments Collision energy ramp from 20 to 40 eV during high energy accumulation

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Figure 1. MarkerLynx report for negative Ion LC/MS analysis of total phospholipids isolated from H4IIE cells. Top pane: Table of components and normalized peak areas. Bottom left pane: Scores plot showing class separation. Bottom middle pane: Trend plot for component at m/z 844.6081@9.71min. Bottom right pane: Loadings plot data labeled with m/z retention time pair.



Figure 2. Narrow XIC for component m/z 844.608 @9.71 min. Top trace: Oleic acid treatment, Middle trace: Control treatment, Bottom trace: Palmitic acid treatment.



Figure 4. Top panel: TOF MS spectrum of component *m/z* 844.6083 @ 9.71 min. Bottom panel: TOF MS^E spectrum.



Figure 3. Elemental composition report for component m/z 844.6083 @ 9.71 min.



Figure 5. Elemental composition report for fragment ions from component m/z 844.6083 @ 9.71 min generated by MS^E. The compound is a PS containing oleate (C18: ω 1) and behenate (C22: ω 0) acyl chains.

| | | | Peak Area | | | Peak Area Ratios | | | |
|--------------|-----------------|---------------|------------|---------|------------------|------------------|----------------|--|-------|
| Phospholipid | Ret. Time Mass | Palmitic Acid | Oleic Acid | Control | Palmitic/Control | Oleic/Control | Palmitic/Oleic | Formula | error |
| PI C36:ω 4 | 6.5671_ 858.523 | 103.62 | 6.41 | 10.80 | 959 | | 1616 | C ₄₅ H ₇₈ O ₁₃ P | 2.8 |
| PI C34:ω 0 | 7.3831_ 835.536 | 153.67 | 12.50 | 49.80 | 309 | | 1229 | C ₄₃ H ₈₀ O ₁₃ P | 3.2 |
| PS C34:ω 0 | 7.4679_ 760.515 | 100.03 | 24.67 | 15.28 | 655 | | 405 | C40H75NO10P | 2.5 |
| PE C36:ω 6 | 7.5368_ 736.5 | 52.30 | 6.75 | 21.18 | 247 | | 775 | $C_{41}H_{71}NO_8P$ | 4.1 |
| PS C40:ω 4 | 7.582_ 838.564 | 110.01 | 12.87 | 36.45 | 302 | | 855 | C ₄₆ H ₈₁ NO ₁₀ P | 5.7 |
| PI C36:ω 2 | 7.7355_ 861.549 | 51.69 | 172.84 | 79.57 | | | | C ₄₅ H ₈₂ O ₁₃ P | 2.7 |
| PS C36:ω 2 | 7.7738_ 786.527 | 24.64 | 59.12 | 15.70 | | 376 | 42 | C42H77NO10P | 3.9 |
| PI C38:ω 4 | 7.818_ 885.551 | 395.23 | 343.93 | 243.25 | 162 | | | $C_{47}H_{82}O_{13}P$ | 3.3 |
| PS C36:ω 0 | 8.0265 790.562 | 544.69 | 229.31 | 318.94 | 171 | 72 | 238 | C ₄₂ H ₈₁ NO ₁₀ P | 1.3 |
| PS C35:ω 0 | 8.0278_ 776.547 | 114.18 | 51.75 | 59.30 | 193 | | 221 | C ₄₁ H ₇₉ NO ₁₀ P | 1.3 |
| PS C42:ω 3 | 8.1788_ 864.578 | 93.67 | 69.88 | 82.73 | | | | C48H83NO10P | 0.3 |
| PS C36:ω 0 | 8.2756_ 790.562 | 274.09 | 17.83 | 76.21 | 360 | 23 | 1537 | C ₄₂ H ₈₁ NO ₁₀ P | -0.1 |
| PE C34:ω 2 | 8.3459_ 714.51 | 65.95 | 61.01 | 68.92 | | | | C39H73NO8P | -1.5 |
| PE C36:ω 4 | 8.3648 738.508 | 137.42 | 108.39 | 123.26 | | | | C ₄₁ H ₇₃ NO ₈ P | -1.9 |
| PS C37:ω 1 | 8.4191_ 802.562 | 48.51 | 81.33 | 60.64 | | 134 | 60 | C ₄₃ H ₈₁ NO ₁₀ P | 1.9 |
| PS C38:ω 1 | 8.4287_ 816.578 | 213.18 | 348.99 | 297.37 | 72 | | 61 | C44H83NO10P | -0.2 |
| PS C40:ω 3 | 8.431_ 840.578 | 200.34 | 64.88 | 67.07 | 299 | | 309 | C46H83NO10P | -0.1 |
| PS C39:ω 3 | 8.4311_ 826.562 | 39.32 | 14.85 | 14.18 | 277 | | 265 | C45H81NO10P | -1.9 |
| PS C38:ω 1 | 8.6363_ 816.574 | 59.05 | 54.29 | 59.34 | | | | C44H83NO10P | -2.4 |
| PI C38:ω 3 | 8.6593 887.566 | 75.18 | 154.84 | 95.78 | | 162 | 49 | $C_{47}H_{84}O_{13}P$ | 3.3 |
| PS C37:ω 0 | 8.7082_ 804.578 | 114.33 | 67.35 | 100.22 | | | 170 | C ₄₃ H ₈₃ NO ₁₀ P | 0.6 |
| PS C36:ω 0 | 8.7613_ 790.549 | 9.01 | 40.66 | 18.45 | | | | C ₄₂ H ₈₁ NO ₁₀ P | 1.4 |
| PE C38:ω 3 | 8.7788_ 764.524 | 76.68 | 129.49 | 89.00 | | 145 | 59 | C43H75NO8P | 1.8 |
| PS C40:ω 2 | 8.8204_ 842.59 | 99.29 | 63.02 | 57.75 | 172 | | | C46H85NO10P | 4.4 |
| PS C35:ω 1 | 8.9861_ 778.561 | 113.29 | 0.00 | 8.67 | 1307 | | | C ₄₁ H ₈₁ NO ₁₀ P | 1.8 |
| PS C36:ω 1 | 8.9962_ 792.577 | 468.38 | 12.49 | 55.20 | 849 | 23 | 3750 | C ₄₂ H ₈₃ NO ₁₀ P | 3.8 |
| PS C39:ω 1 | 9.0435_ 830.59 | 38.31 | 75.47 | 81.20 | 47 | | 51 | C45H85NO10P | 3.7 |
| PE C34:ω 1 | 9.3262_716.522 | 106.95 | 82.80 | 87.12 | | | | C39H75NO8P | 0.1 |
| PS C37:ω 0 | 9.3789_ 804.578 | 157.24 | 122.86 | 129.69 | 121 | | 128 | C43H83NO10P | -3 |
| PS C38:ω 0 | 9.3798_ 818.593 | 528.37 | 460.42 | 454.46 | 116 | | 115 | C44H85NO10P | 1.1 |
| PE C40:ω 5 | 9.4424_ 790.54 | 38.49 | 71.17 | 45.77 | | 155 | | C45H77NO8P | 2.1 |
| PS C44:ω 5 | 9.4955_ 892.61 | 58.05 | 34.60 | 40.33 | 144 | 86 | 168 | C ₅₀ H ₈₇ NO ₁₀ P | -0.3 |
| PS C40:ω 1 | 9.5613_ 844.612 | 76.00 | 68.61 | 57.24 | | | | C46H87NO10P | 3.8 |
| PE C36:ω 2 | 9.6235_ 742.54 | 72.54 | 248.79 | 100.82 | 72 | 247 | 29 | C41H77NO8P | 2.3 |
| PS C40:ω 1 | 9.7152_ 844.608 | 137.18 | 614.38 | 233.25 | 59 | 263 | 22 | C46H87NO10P | 3.3 |
| PS C39:ω 1 | 9.7188_ 830.592 | 47.03 | 218.82 | 62.61 | 75 | 349 | 21 | C45H85NO10P | 5.1 |
| PE C38:ω 4 | 9.7203_ 766.54 | 300.09 | 258.48 | 199.41 | 150 | | | C ₄₃ H ₇₇ NO ₈ P | 4.3 |
| PS C42:ω 3 | 9.7621_ 868.608 | 39.07 | 28.38 | 32.72 | | | | C48H87NO10P | 4.5 |
| PE C40:ω 5 | 9.7667_792.555 | 71.14 | 24.01 | 42.74 | 166 | | 296 | C45H79NO8P | 5 |
| PS C42:ω 2 | 10.164_ 870.629 | 57.23 | 23.94 | 34.82 | 164 | | 239 | C48H89NO10P | 4.7 |
| PS C41:ω 1 | 10.271_ 858.625 | 19.24 | 33.80 | 33.41 | | | 57 | C47H89NO10P | 3.7 |
| PE C38:ω 3 | 10.555_ 768.556 | 61.39 | 64.50 | 42.47 | 145 | 152 | | C43H79NO8P | 2 |
| PE C36:ω 1 | 10.604_ 744.554 | 163.39 | 166.97 | 150.29 | | | | C ₄₁ H ₇₉ NO ₈ P | 5.9 |
| PS C40:ω 0 | 10.658_ 846.623 | 353.63 | 303.46 | 316.18 | 112 | | 117 | C46H89NO10P | 0 |
| PS C39:ω 0 | 10.665 832.609 | 140.42 | 114.10 | 123.72 | | | 123 | C45H87NO10P | 4.4 |
| PS C42:ω 1 | 10.875_ 872.639 | 82.05 | 134.20 | 76.91 | | 174 | 61 | C ₄₈ H ₉₁ NO ₁₀ P | -0.8 |
| | | | | | | | | StdDev (ppm) | 2.3 |

(95% confidence level (Students t > 2.78)) are expressed as a ratio of mean peak areas* 100.



Figure 6. Top panel: XIC of m/z 184 from LC/MS^E analysis of phospholipids. The m/z 184 ion is specific to phosphatidylcholine and sphingomyelin. Bottom panel: ESI +ve LC/MS BPI chromatogram.

Table 1. Identification of phospholipids observed in negative ion ESI/LC/MS analysis of H4IIE cells. Phospholipids are identified by class, acyl chain carbon number and degree of unsaturation. Significant peak area differences



Figure 7. Elemental composition report for MS^t spectrum at 8.11 minutes. The phospholipid is PC C32:ω 1 (m/z 732.5560).

Data Visulization with SIMCA-P





Figure 8. SIMCA-P orthogonal Partial Least Squares (o-PLS) plots for control and Palmitic acid treatment groups. Right panel: Scores plot. Left Panel: Loadings plot.



Figure 9. O-PLS 'S' Plot for Palmitic acid treatment vs. Control group. This plot highlights the components responsible for the group separation.

CONCLUSIONS

- Clear separation of Palmitic acid treatment group from **Control and Oleic acid treatment groups**
- Identification of phospholipids that cause class separation
- Use of MS^E to determine phospholipid acyl chain composition
- Use of positive ion MS^E to identify and characterize PC and SM phospholipids

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